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CENTRAL INTELLIGENCE AGENCY	REPORT
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COUNTRY USSR / International	CASE DATE DIS
SUBJECT Work on the Chemistry of Collagen and Connective Tissue	NO. OF PAGES 3
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DATE OF INFORMATION 25X	
1. The chemistry of connective tissue is of considerable i cine. During World War II when natural gut was not ava quantity to meet the demand, satisfactory sutures were collagen. Tubular collagen was successfully used to co and pieces of collagen in various shapes were used in second pieces of collagen in various shapes were used in second speed at which the material is absorbed by the body. 2. In the period 1947-1950 a group of Soviet biochemists deposite work on the chemistry of the last of the second states.	ilable in sufficient produced from reconstituted nnect severed nerve ends, erious wound operations. was subjected, the degree could be controlled.
basic work on the chemistry of collagenous tissue	
M. P. <u>Chernikov</u> N. N. <u>Dobbert</u> P. A. <u>Finogen</u> ov	
(S. Y.?) Frenkel A. S. Konikova V. N. Orekhovich	
K. D. <u>Orekhovich</u> N. Y e. <u>Plotnikova</u>	
A. D. Speranskiy A. A. Tustanovskiy	· ·
have published papers on the subject. In general these in which they extracted collagen from animal tissue by mafter getting the collagen in solution they reconstitute tic fiber. They were primarily interested in the chemis distribution in animal tissue, and its connection to the Results of their work were published by the Academy of M journal "Biochimia" and elsewhere, including the following Chernikov, M. P. "Amino Acid Content of Ox Procoll Nauk, U.S.S.R. 67, 345-7, 1949. "The Amino Acid Properties of Boy	means of acidic buffers. Ed the collagen as synthe- stry of the substance, its e precursor of animal cells. Medical Sciences in the State of the substance, its edical Sciences in the substance, its edi
of the Academy of Sciences of the USSR, Orekhovich, K.D. "The Procollagen Content of Skin Ages". Akademiia Nauk USSR Doklady, 71: "The Procollagen of Hide". Biokh	IXVII, 2, p. 345-9. in Animals of Different 521, 1950.

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Orekhovich, V. N., Konikova, A. S., Orekhovich, K. D., and Dobbert, N. N.

*Concerning the Metabolic Turnover Rates of Various Organ and
Tissue Proteins*. Akademiia Nauk (SSR Doklady, 71:105, 1950.

Bresler, S. I., Finegenov, P. A., and Frenkel, S. Y. A discussion of the
structure of the Magromolecule of Procollagent. Reports of the
Academy of Sciences of the U.S.S.R., IXXII (72), 3, pp. 555-8, 1950.

Tustanovskiy, A. A. "Concerning the Proteins of Skin". Biokhimiia, 12 (4):
285-290, 1947.

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- 3. Two other Soviets, S. L. <u>Pupko</u> and A. L. <u>Zaydes</u> have also written on electron-microscopic investigations of collagen published in proceedings of the Academy of Sciences:
 - Zaydes, A. L. and Pupke, S. L. "Electron-Microscopic Investigation of the Effects of Alkalis and Pancreatin on Collagen". Akademiia Nauk USSR Doklady, Vol. 73: 991, 1950.

 "The Electron-Microscopic Examination of Collagen". Reports of the Acad. of Sci. of USSR, LXV, 2, pp. 227-8, 1949.

 "The Electron-Microscopic Examination of Collagen Using the Replica Technique". Akademiia Nauk USSR Doklady, Vol. 73: 379, 1950.

25X1A

The fact that they are members of the Central Research Institute of the Leather Footwear Industries leaves little doubt as to the Soviet belief in the future of reconstituted collagen.

- 4. US scientists reviewing this work found the Soviet conclusions particularly interesting. For the past three years 49-52 a group of US scientists has been working on the ultrastructure of connective tissue based on the work of the above named Soviet scientists. The US group also hopes to gain information on the formation and nature of collagen and determine its molecular structure. To date attempts of US scientists to make contact, through the Soviet Academy of Science, with the Soviet collagen specialists have been unsuccessful. The Soviets also have never mentioned US work in the field; so it is hard to assay Soviet progress.
- 5. In these days of the successful use of such agents as cortisone, many people are getting on the bandwagon to investigate connective tissue, and the field is becoming increasingly more popular. Results of this research will find application in medicine and may give insight into such phenomena as wound healing and aging. The production of synthetic fiber and synthetic leather as a result of this research is quite possible.

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8. This side of the Iron Curtain, the British are probably the leaders in the field.

Germany and Scandinavia are also very active in tissue ultrastructure. There seems to be some work going on in Italy.

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Bresler, 5.4., P. A. Finogenov, and S.Y. Frenkel. A discussion of the structure of the Macromolecule of procollagen. Reports of the Academy of Sciences of the U.S.S.R., LXXII (72),3, pp. 555-8, 1950.

In 1947 V. N. Orekhovich and A. A. Tustanovsky (1) isolated a new crystalline albumin from the skins of animals, which they called procollagen, and studied its properties in detail (2-8). Aside from the purely biological problems which arose because of this discovery, the new albumin presented much interest from the standpoint of the physical-chemistry of macromolecules. By all external appearances it was similar to fibrous albumins, since, for instance, it gave a solution with high viscosity. On the other hand, a strong tendency to form crystals made it comparable to the globular albumins.

1. We investigated the procollagen from the skin of rats. The sedimentation constant S of the albumin was measured in the ultracentrifuge in concentrations of 0.045 to 0.45%. We also found the diffusion constant D at several concentrations, the specific volume and the viscosity. The experiments in the ultracentrifuge were carried on at 60,000 RPM, giving an acceleration of 250,000 g., and the sedimentation velocity was followed by the scale method of Lamma. The diffusion measurements were made on an apparatus provided with automatic recording devices over the period of two to three days.

For every experiment a fresh solution of procollagen was prepared in 0.025 citrate buffer at a pH of 3.0 (this pH is the most stable one for albumin). Afterwards the solution was filtered to remove a small amount of flocculent precipitate, using a No. 4 glass filter, upon which it appeared perfectly transparent. The percentage of procollagen was determined from the nitrogen content.



2. Who first thereigh investigation by us of We proper des of procollegen disclosed that of it did not become described, this protein was
really a monedisperse allowin, and therefore that it really belonged to the
group of globular proteins. Fig. 1 shows the sedimentation of procedinger
in the ultracentrifuge (albumin cone. of 0.21%). In Fig. 2 in presented
the variation of the sedimentation constant 3, remultized as usual to 20° 3,
and pure water, as a function of the concentration of the allowin. The increase of the sedimentation costant with a decrease in concentration is ereplained by the fact that large elengate molecules do not rows independently
of each other in concentrated adultions.

The measurement of the diffusion constant of procellagen presents contain difficulties, for when the albumin is maintained for some time at 25° C. it starts to denature and coagulate. Nevertheless, under high enough concentrations (above 0.5%) the error caused by this deformation is negligibly small and allows the calculation of the coefficient of diffusion with gratifying accuracy. In the concentration interval between 0.3 and 0.5% the diffusion constant has an average value of 2.2h x 10°7 cm²/tec. (adjusted also to pure water and 20° C.). One can accure that an average value of 10 occurs at the center of the interval, i.e. at 0.4%.

With S and D known it is possible to calculate the molecular usignt of the procollagen by the formula of Swedbergs

(3)

Here S is the sedimentation constant, i.e. the relocity of the macromolecules divided by the acceleration; A is the molecular weight; I is the specific reluce:

C is the density of water at 20° C. (.3982); I is the diffusion constant;
T the absolute temperature (293° K.); and R the universal gas constant (8.313° 107 erg/degrees).

The specific volume was measured using a pycnometer and turned out (at 20° C) V = $0.720 \div 0.4005$.

By using the value at 0.1% conc. From Figure 2, $S = (1.8 \pm 0.05) \times 10^{-13}$ one obtains for the molecular weight of procollagen $M = 70,000 \pm 3500$. This allows one to compare procollagen to the large group of globular albumins near the weight of 70,000, which corresponds to four of the elementary units of Svedberg (17,500).

3. Now we consider the degree of asymmetry of the albumin. For this ourpose we use the value of S extrapolated back to zero concentration of albumin, i.e. $S = 3 \cdot 10^{-13}$ (Fig. 2). From the formula of sedimentations

$$S = \frac{M(1 - eV)}{f} \tag{2}$$

we find the frictional force f. This frictional force is related to the movement of the macromolecules in an infinitely dilute solution, i.e. when it does not depend on their interaction. If the macromolecules are spheres one can apply Stokes law $f_0 = 6\pi$ ya, where a $\frac{3}{2}$ $\frac{2}{12}$ radius of the sphere (N is Avogadro's number).

For this case we make use of the empirically found relationships

(3)

If this equation (and it is exactly of the same magnitude as ob-

explained by the asymmetry of the sphere it can be used to relate the formula for frictional forces to the dimensions of a prolate ellipsoid:

$$\frac{f}{f_0} = \frac{\sqrt{1 - \lambda^2}}{\lambda^2} \int_{\Omega} \int_{\Omega} \frac{1}{\lambda^2} \frac$$

Using our data the ratio of two semi-exes becomes:

Consequently, the molecule of procollagen presents a cylinder with a length about 20 times its maximum diameter. Knowing the volume of the molecule $v = \frac{MV}{N}$, we find that the diameter $d = 16.7 \, \text{Å}$, and the length $L = 380 \, \text{Å}$.

Since the approximate amino acid composition of the macromolecule is known (8) we can calculate the average molecular weight of the residues, m = 117, and therefore the degree of polymerization of the procollagen is the length of a single peptide bond equals about h Å, the length of the whole polymeptide chain in procollagen must be 2400 Å, that is 6.25 times the length of the length L of the macromolecule. This undoubtedly means that the polypeptide chain is coiled in the macromolecule of the albumin.

4. In conclusion we will discuss the morphological reactions of a solution of procollagen under the action of salts. It was known that the presence of salt directly affects a solution of procollagen, changing, in particular, its viscosity. We examined in the ultracentrifuge a solution of procollagen in 0.32 M NaCl at a ph of 3.0.

The result was that next to the original peak of the procollagen appeared two new peaks, having a somewhat higher sedimentation rate (Fig. 3). This indicated the appearance of "dimers" because of the association of globules. Evidently this association can take place by different paths. The simplest variation of this association consists in the elongation of the globule by

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two (association in length). Using formulas (2) and (3) easily shows what will happen in this case.

Equation (3) may be written as:

$$\left(\frac{S}{F_0}\right)_1 = \frac{const}{S_1} \frac{77}{S_1}$$

For the dimer we have the corresponding

$$\left(\frac{f}{f_0}\right)_2 = \frac{\text{const (arn)}^{e_{i_0}}}{S_c}$$

Therefore

$$\frac{S_a}{S_i} = a^{a/3} \left(\frac{\xi}{\xi_i} \right)_i / \left(\frac{\xi}{\xi_i} \right)_a$$

(5)

In our case of linear association $(\frac{5}{5})_2 \approx 2.78$ (the length of the cylinder is doubled while the radius remains constant). This would give theoretically $\frac{5}{5} \approx 1.21$. In actuality we obtained $S_2 \approx 3.083 \times 10^{-13}$, $S_1 \approx 2.558 \times 10^{-13}$, and therefore $S_2/S_1 \approx 1.20$, which constitutes an ideal correlation.

Definitely interesting is the second peak, a sedimentation with the values of $S_3 = 3.91 \times 10^{-13}$, $S_3/S_1 = 1.53$, which is almost equal to $2^{2/3}$ (1.59). This corresponds to an association of the two macromolecules along the length of the sides of a forming cylinder. In this case the semi-minor axis of the ellipsoid, as the length, stays constant, and the relation ℓ/ℓ_0 of the dimens is approximately equal to that of ordinary nolecules.

This experiment demonstrates that under the action of salts, as is known, the ionic attraction between primary groups of the albumin is decreased (at a pN of 3.0 the dissociation of the acid groups is mainly supressed). We also see that molecules of procollagen combine in pairs, forming two types of dimers - linear and lateral.

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Translated by Walter R. Stahl November 20, 1950

Chernikov, M. P. Amino acid content of ox procollagen. Doklady Akad. Nauk. U.S.S.R. 67, 345-7, 1949. (Presented by the Academician A. D. Speransky, May 5, 1949).

For the past three years U. N. Orekhovich and his collaborators have been conducting a thorough study of the procollagen (1-12) of proteins which were isolated in the laboratory in crystalline form from the skin and internal organs of humans and various animals.

Along with the study of the biological and physical chemical properties and the investigation of the prevalence of protein in the tissues of various animals (13,14) work was also conducted in the investigation of the qualitative (N. E. Plotnikov) and quantitative amino acid composition of the given protein. Some of the results of the chemical investigation of procollagen are set forth in this paper.

The procollagen was obtained from the hide of an ox immediately after skinning. The hide was washed, shaved, the fat and cellular tissues carefully removed, after which the hide was washed again in a current of running water and was ground in a meat chopper after having been cut up into small pieces. The resulting paste was used for the isolation of protein, globulin and procollagen.

After extracting the albumen and globulin with 1/15 M Na₂PO₄ the paste was washed with a citrate buffer, pH = 3.5.

The procollagen was extracted with a triple quantity of citrate buffer in the course of 3-4 days at a temperature of 42°. Thymol and toluene were used in all cases for preservation.

The amorphous preparation of procollagen was obtained from the citrate extract by saturating it with dry NaCl up to 10%. The precipitate of albumen settled in the form of white flakes which were collected on the filter. To cleanse the procollagen from concomitant albumen, it was washed on the filter many times with a 5% solution of NaCl. To remove salts the procollagen was made turbid in a 15-20% solution of acetone with a subsequent centrifuging.

This operation was repeated to a negative reaction for chloride ions and a negative Millon's reaction, since the absence of tyrosine indicates the purity of procollagen compounds. Beyond that the protein was dried by the regular method with acetone and ether.

The resulting compound of the ox procollagen contained 0.47% ash, 17% nitrogen, 49% carbon and 7% hydrogen.

To determine tryptophane, tyrosine and phenylalanine, an alkaline hydrolysis was carried out with 5 N solution NaOH in the course of 6 hours by light boiling over a sand bath. The alkali was neutralized with a 14 N solution of H₂SO₄. The hydrolyzed substance was brought to a definite volume and then filtered.

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To determine the remainder of the amino acids an acid hydrolysis was carried out over a sand bath with 6 N solution of HCl in the course of 36 hours. The hydrolyzed substance was filtered after dilution. The hydrochloric acid was removed in a vacuum after which the hydrolyzed substance was brought to a definite volume.

For the analysis of amino acids we used mainly specific methods which do not require any preliminary separation of amino acids. The methods were checked on hydrolyzed casein or gelatin. In all instances (except proline and oxyproline) we obtained fully satisfactory results.

Tryptophane (15, tyrosine (16), phenylalanine (17), methionine (18), cystine and cysteine (19), proline and oxyproline (20), were determined colorimetrically. Glycine and alanine by acidification with ninhydrin and the ensuing colorimetric determination of the resulting aldehydes (21,22). Arginine, histidine, lysine, amino succinic and glutamic amino acids were determined by the enzyme method (23,24) (in the laboratory of B. E. Zbarsky).

The determinations were made on a Schtufenphotometer or on a Specker absorptiometer.

The results of our investigations are tabulated in Table I. This table also shows the amino acid composition of collagen and gelatin (according to published data).

Table I

	In %	Dry Ash-free P	rotein
Amino Acide	Collagen (25)	Procollagen	Gelatin (26)
Tryptophane Tyrosine Phenylalanine Arginine Lysine Histidine Amino succinic acid Glutamic acid Cystine	0.4 4.28 4.58 0.33	0.0 0.0 2.3 9.2 4.6 2.9 5.2 11.0	0.0 0.44 2.2 8.0 4.1 0.79 6.7
Cysteine) Methionine Glycine Alanine Proline Oxyproline	0.0 0.08 26.2 9.5 15.1 14.0	0.0 0.66 28.0 9.5 (20)	0.07 0.61* 25.5 5.7 19.7 14.4

^{*}In accordance with our data, in gelatin of the hide of a calf.

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By comparing procollagen and collagen in their amino acid compositions it can be seen that they contain equal quantities of lysine, glutamic acid, alanine, arginine, amino succinic acid, methionine and glycine and neither of the two contains tryptophane, cystine and cysteine.

Procollagen and collagen differ in their contents of phenylalnine, histidine, proline and oxypoline. Procollagen, unlike collagen, does not contain tyrosine.

Procollagen also differs in its amino acid composition from gelatin as is seen from the data presented in the above table. They contain different quantities of histidine, arginine, amino succinic acid, proline and oxygroline, and, furthermore, gelatin contains tyrosine while procollagen does not.

Thus, on the basis of our data, it can be said that procollagen in its amino acid composition is close to collagen and gelatin but is not identical to them and appears as an individual protein with connective tissues of the collagen type.

I am deeply grateful to Prof. V. N. Orekhovich for his valuable instructions and guidance in the performance of this work.

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THE AMINO ACID PROPERTIES OF BOVINE PROCOLLAGEN

M. P. Chernikov

(Reports of the Academy of Sciences of the USSR, LXVII, 2, p. 345-9)

V. N. Orekhovich and his associates have been conducting a multilateral investigation of the protein, procollagen, during the course of the last three years (1-12). They have obtained this substance from the skin and internal organs of both man and animals, using a product in crystalline form.

Together with the investigation of biological and physio-chemical properties, and also the distribution of the protein in question in different animals (13,14), there has been work done on the qualitative (N.F. Plotnikova) and quantitative amino acids properties of procollagen. Some of the results of the chemical studies of this protein follow.

Procollagen was obtained from cow hide freshly removed from the animal. The skin was washed, the covering of hair removed, the fat and subcutaneous cellular substance eliminated, and the hide then again washed in running tap water. The skin was ground up in a meat chopper after being cut into small sections. The mash so obtained was used to get albumins, globulins, and procollagen.

After the extraction of albumins and globulins with 1/15 M solution of Na₂HPO₁, the mash was washed with a citrate buffer of pN 3.5. The procollagen was removed with a three-fold soaking in citrate buffer over a period of 3-4 days at a temperature of 2°C. In all cases thymol and toluol were used as preservatives.

An amorphous preparation of procollagen is obtained by adding solid NaCl until a 10% solution was obtained. The protein precipitates as a white mass which is collected by filtration. In order to remove associated proteins, the filtrate is washed a number of times with 5% NaCl solution. In order to remove the salt, the procollagen was soaked in 15-20% acetone and subsequently centrifuged. This operation was repeated until one obtained a negative reaction for chloride ions and a negative Mellon's test, since the absence of tyrosine is a good criteria for purity of procollagen. Thereupon the protein was dried in the usual fashion with acetone and ether.

Procollagen prepared in this way has 0.47% ash, 17% nitrogen, 49% carbon, and 7% hydrogen.

For the determination of tryptophane, tyrosine and phenylanaline, the protein was hydrolyzed in 5N NaOH for six hours, with gentle boiling on a sandbath. The base was neutralized with a llN solution of sulfuric acid. The hydrolysate was then brought to a definite volume and filtered.

In order to determine the other amino acids, the procollagen was hydrolyzed in 6N HCl for 36 hours, with heating on a sand bath. After dilution the hydrolysate was filtered. The hydrochloric acid was removed in a vacuum and the volume adjusted to some definite amount.

In order to analyze the amino acids we used the accepted specific methods which do not require a preliminary separation of component amino acids. The process was checked with gelatin or casein. In all cases (with the exception of proline and oxyproline) we obtained fully satisfactory results.

Tryptophane (15), tyrosine (16), phenylalanine (17), methionine (18), cystine and cysteine (19), proline and oxyproline (20) were determined color-metrically. Glycine and alanine were found by colorometric measurement of aldehydes following oxidation with ninhydrin (21,22). Arginine, histidine, lysine, aspartic and glutamic acids were determined by enzymatic methods (23,24). (In the laboratory of B.E.Zbarckogo)

In Table I is presented the amino acid composition of procollagen with those of gelatin and collagen, obtained from literature. The figures are percentage of dry, salt-free protein weight.

	Collagen (25)	Procollagen	Gelatins(26)
Tryptophane	0.0%	0.0	0.0
Tyrosine	1.4	0.0	0. hh
Phenylalanine	կ2	2.3	2.2
Arginine	8.8	9.2	8.0
Lysina	4.5	4.6	4.1
Histidine	0.8	2.9	0.79
Aspartic acid	6.3	5.2	6.7
Glutamic acid	11.3	11.0	11.5
Cystins) Cysteine)	0.0	0.0	0.07
Methionine	0.8	0.66	0.61 (out own data)
Glycine	26.2	28.0	25.5
Alanine	9.5	9.5	8.7
Proline	15.1	(20)	19.7
Oxyproline	14.0	(20)	Theh

Upon comparing collagen and procollagen according to amino acid composition we can see the following facts: both contain the same amount of lysine, glutanic acid, alanine, arginine, aspartic acid, methionine, and glycine and both do not contain any tryptophane, cystine or cysteine.

Procollagen and collagen differ in their contents of phenylalanine, histidine, proline and oxyproline. Procollagen, in distinction to collagen, does not contain any tyrosine.

A similar analysis can be done for gelatin. We thus see that though procellagen is similar to collagen and gelatin in amino acid composition, it is not identical. We may conclude it is a unique connective tissue protein.

At the Institute of Biological and Medical Chemistry Academy of Sciences of the U.S.S.R.

presented 5/4/49

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Trans. W. Stahl 4/6/51

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Approved For Release 2003/01/29 : CIA-RDP80-00926A005400050022-5 K. D. OREKHOVICH

(Akademiia Nuak USSR Doklady, 71:521, 1950)

THE PROCOLLAGEN CONTENT OF SKIN IN ANIMALS OF DIFFERENT AGES

A new group of connective tissue proteins - procollagens, which were discovered by V. Orekhovich and A. Tustanovski, has been thoroughly investigated during the course of the last year (1-13). At the present time, we know the chemical composition, chemical and physio-chemical properties of procollagen. There has been much work done on the biological significance of this protein.

In this paper we present some observations based on the last series of experiments devoted to procollagen. The investigations deal with the content of procollagen in the skin of guinea pigs of various ages (from ten days to one year and more).

From the skin of normal, healthy animals we extracted procollagen by the method of A. Tustanovich (3) and determined the amount of this protein by comparing it with the weight of the dry, degreased skin; as well as by comparison to the mass of skin protein--collagen. The akin, removed immediately after killing the animal, was freed from hair, subcutaneous fat and cells, ground-up and treated by the methods given previously (204). The extraction of the protein was repeated five times. The procollagen thus obtained was dried to constant weight. The results are given in table 1.

As we see from table 1, guinea pigs at a young age (from 10 day to 5-6 months) have an amount of procollagen which varies between 7-103. More mature animals (7-8 months) show a drop to 3-4%, while the old guinea pigs (8 months and older) have only 1-2% of orocollagen.

Table 1

Age	Weight when killed (g)	% Pro- collagen	Age	Weight	% Pro- collagen
10 days 20 " 2-6 months " " " " " " " " " " " " " " " " " " "	80-117 120-170 262 287 3347 3467 367 377 3400 402 4454 535	6.7* 7.320 7.554 7.04 10.454 10.752 10.454 10.752	7-8 months " " " " " " " " " " " " " " " " " "	670 715 740 760	4.0970.0.065830681

In order to answer the objection that our data depends on the different extraction conditions optimum for older and younger animals, we conducted the following experiment.

We took the skin from five animals weighing 300-400 grams and from two guinea pigs of 725 and 827 g. The mash obtained from these two groups of animals was divided into eight portions. Each aliquot was covered with buffer solution of the following oH: 1.5, 2.0, 3.0, 3.6, 3.95, 4.12, 4.47, 5.02. Procollagen was obtained from the extracts. We found (see Table 2) that the results were the same regardless of the extraction used, and that they also corresponded to earlier conclusions, i.e., that the skin of older enimals has several times less procollagen than that of younger ones, although the optimum extraction conditions vary for the two groups.

On the basis of the facts presented above, we can be confident of the fact that as an animal gets older the percentage of procollagen in its skin decreases. It is notable, that in old animals the process of formation of new collagen fibers is also slower. There is reason to suspect that there is some relationship between the formation of collagen and the concentration of procollagen within the skin.

Teble 2

THE AMOUNT OF PROCOLLAGEN OBTAINED FROM THE SKINS GUINEA PIGS UNDER DIFFERENT EXTRACTION CONDITIONS

Weight (g)	1 =	2.0	3.0	он во 3-6	lution	h.12	և "47	5.02
Weight (g) 300-400 725, 827	11.6	14.3	17.1	12.3	4.2	7.2 4.1	12.0 3.4	8.8 2.2

Institute of Biological and Medical Chemistry of the Academy of Medical Science of the USSR

Presented by: 7/1/1949

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We have isolated and obtained out of the hides of animals, albumin in crystalline form which evidently belongs to a special group of connective tissue albumin. There are some reasons to assume that it is a biochemical predecessor of collagen, and we therefore named it procollagen (1-4). In this article are given the data of the prevalence of procollagen in the animal world and some facts about the mature of this albumin.

Material and Methods

For the purpose of experimentation animals of various classes of vertebrates and invertebrates were taken. Of the mammals rabbits, rate, calves, dogs and cats were taken; of birds - hens; of reptiles - turtles and grass snakes; of amphibians - frogs; of fish - pike perch. Besides that, tests were made (for content of procollagen) with human skin and tissues of various species of invertebrate animals.

The hides of animals were taken immediately following their slaughter and the removal of their blood. The outer coverings (wool, scales, feather and horny membranes), hypodermic tissue, fat, etc. were carefully removed and the remaining material was ground up. The resulting paste was used for the purpose of isolation of albumins, globulins and procollagen (crystalline and amorphous).

The albumins and globulins of the animal hides were isolated from the paste by means of five times the volume 0.3 M Na₂HPO_{μ} within 24-36 hrs. at $^{+}2^{2}$. The globulins were isolated by adding an equal volume of a saturated solution of $(NH_{\mu})_{2}SO_{\mu}$. By adding $(NH_{\mu})_{2}SO_{\mu}$ to saturation, the albumins were isolated from the filtrate.

The crystalline procollager was isolated by the following method. After extraction of the albumins and globulins by means of the phosphate, the remaining paste was washed once or twice with small batches of citrate buffer pH = 4.0. To the washed out paste was added citrate buffer (pH = 4.0) five times the amount by volume (the weight of the paste). This mixture was left at a temperature of (+1°, +2° C) for 24-36 hours. After filtration a transparent viscous solution containing procollagen was obtained. This filtrate was put in collodion or cellophane bags for dialysis under running water or 0.01 M Ma2HPO4. In 24 hours crystals of procollagen in the shape of long needles were precipitated. The crystalline procollagen was collected on a filter and dried with filter paper. The moist crystals were kept cool. To obtain dry albumin the crystalline procollagen was desiccated with alcohol and ether and then dried to a constant weight at 104-105° or the

2.

dry crystals were obtained by the method previously published by us (1).

Amorphous preparation of procollagen we obtained by the following method. To the citrate extract of the hide paste was added an equal volume of a 10% solution of sodium chloride. The procollagen was isolated from the solution. The precipitate of albumin was collected on the filter and washed several times with a 4-5% solution of sodium chloride. The moist procollagen was kept in the cold.

The ultraviolet spectra of the absorption of the albumin solutions were studied with the aid of a Smith and Hensh quartz spectrograph (medium dispersion) by the method of the threshold of blackening. The source of light was a hydrogen lamp of low voltage of the GOI system.

The procollagen was boiled with crystalline peosin (2), trypsin and chymotrypsin (prepared by method of Kunitz and Northrop) and preparations of papain and patential (an acid glycerine extract from the liver of a rabbit)

Results of Investigations

The Occurrence of Procollagen

We encounter procollagen in the skins of all classes of vertebrate animals.

a. Mammals. We found procollagen in the skin of all the animals we examined (rats, rabbits, calves, dogs and cats). The amount as well as the conditions of its extraction and crystallization vary with different animals. The age of the animal is evidently important. The younger the animal the easier it is to extract the albumin and the easier it crystallizes.

The skin of rabbits contains about 4% of procollagem (to the dry weight of the skin). The albumin crystallizes out of the extracts more completely and much faster with a citrate buffer solution of an initial pH = 3.8. The crystals are in the shape of needles with a maximum length of 260 µ (see diag. 1). Out of the skin of a dog the procollagen is isolated in a much lesser amount, namely 0.6%. The crystals form during dialysis from the extracts with citrate and buffer solutions with an initial pH of from 3.0 to 4.0. The maximum length of the crystals is 130 µ. Crystal procollagen was also isolated from the skin of a cat. The maximum length of the crystals is 130 µ. The albumin crystallizes much faster in the process of dialysis under 0.01 M Na₂PO₁, the crystallization proceeds much slower in the dialysis under running water. Procollagen becomes isolated very easily

3.

out of the hide of a newly born calf. The albumin is crystallized from the extracts with citrate buffer solutions with an initial pH of from 2.0 to 5.4. The yield of albumin is 0.86%. The maximum length of crystal is 195 μ (see diag. 2)

- b. Birds. The albumin was isolated from the skin of chicks about 1.5 months old. The crystallized albumin is precipitated from the extracts with buffer solutions of an initial pH 2.7-5.5
- c. Reotiles. The albumin was isolated from the skin of a Middle Asiatic desert turtle. The procollagen of this reotile discloses the peculiarity that the albumin crystals can be obtained only if the pH of the citrate buffer solution (used for extraction) does not exceed 3.0 and the dialysis is carried out under 0.01 M Na₂HPO₄ but not under running water. The maximum length of crystals is 52 µ. So far we have not succeeded in isolating the procollagen from the skin of a grass snake.
- d. Amohibians. Crystalline procollagen in only very small quantities can be obtained from the skin of frogs (Rana temporaria). The dialysis of the extracts is better under a solution of Na2PO $_{\mu}$. Maximum length of crystals is 65 μ .
- e. Fish. The skin of fish, seemingly, contains the largest amount of procollagen. 2.5% of procollagen (dry albumin with dry weight of skin) was obtained from the skin of a pike-perch. The albumin passes into the solution very easily and crystallizes in the dislysis under running water as easily as under 0.01 M Na₂HPO_{μ}. The albumin crystals precipitate from all the extracts with citrate buffer solutions with a pH of from 3.5 to 5.5. The maximum length of crystals is 90 μ . So far we failed to isolate the procollagen from the tissues of the invertebrates. It is possible that the modes of extraction of this albumin have to be different from the ones developed by us in the case of vertebrates.

The Nature of Procollagen

Procollagen is a globular albumin soluble in acidified water and not soluble in neutral and weak alkaline medium. In concentrated solutions it has a very high viscosity. It precipitates from solution even in the presence of small concentration of sodium chloride (5%) and other neutral salts. In amino acidic composition it approaches collagen and gelatin, but in solubility and salting out and other properties it differs from them. During boiling of the solutions or during suspension, procollagen turns to gelatin.

Below we will describe certain facts which characterize the nature of this albumin.

4.

- a. <u>Ultraviolet Abscrotion Spectra</u>. We have made a study of the ultraviolet absorption spectra of solutions of procollagen, albumin and globuline of skin and gelatin.
 - Diag. 3. Ultraviolet absorption spectra of solution of albumin and globulin of skin. The unbroken line is that of albumin and the dotted line is that of globulin.

For analysis we precised 1, 3, 4, 6 and 10% acidified water solutions of procollagen pH of solutions from 2 to 5), 1, 3, 5, 6 and 10% water solutions of gelatin from the hide of a calf (pH of solutions from 2 to 7). Since the majority of the above indicated solutions at room temperature form a jelly, it was necessary to heat the solutions to 30-35° in order to fill the cuvette of the spectrograph. Albumins and globulins of the skin were used in the form of 0.75% salt solutions of albumins (pH of solution - 6.5-7.0).

Solutions of albumins and globuline of skin give a typical albumin spectrum with a maximum absorption of about 2,800 A(see diag. 3) which is defined in these albumins by the presence of considerable quantities of tyrosine and tryptophane. In the spectra of procollagen solutions, the characteristic maximal absorption is absent and only in the spectra of concentrated (over 3%) solutions of albumin there are 5 absorption lines, three of which have a width of 30 A each and 2-20 A each. These lines lie between 2570 and 2,600 A; 2,640-2,660 A; 2,690-2,710 A; 2,760-2,790 A; 2,840-2,870 A (see diag. 4).

In the spectra of 6% and 10% solutions of gelatin obtained from the hide of a calf, we succeeded in detecting only 3 absorption lines 2,570-2,600 Å, 2,640-2,660 Å and 2,680-2,700 Å (diag 5). Comparing the absorption spectra of these two albumins it is seen that the absorption lines for gelatin noted by us are identical with the corresponding lines in the absorption spectra of procollagen. The next 2 lines of absorption which lie in the longer wave section of the ultraviolet we could not detect in the spectra of the gelatin solutions. As is known (5), the characteristics of absorption spectra of gelatin are dependent upon the absence from the albumin of tyrosine and tryptophane and on the presence of phenylalanine. On the basis of the above-mentioned facts we can assert that phenylalanine is also present in procollagen. For the time being we cannot explain on what the characteristic of the absorption spectrum of procollagen depends.

b. The Relationship of Procollagen to Proteinases. We made a study of the digestibility of crystalline procollagen with various proteolytic enzymes. It became clear that this albumin is well digested by tissue proteinases (cateosin and papain) and comparatively poorly by the proteinases of the digestive system

5.

"Procollagen of Hide"

(pensin, trypsin and chymotrypsin) (see table below)

TABLE I

The Intensity of Fermentative Hydrolysis of Procollagen at Various pH Media

•	Increa	se of	Amino	Nitros	en in	24 hrs	, in	specime	n (in	mg.)
Enzymes					g q					
	2.3	3.2	3.67	4.0	4.4	4.5	4.9	5,60	6.0	7.8
papain	1.30	1.40	2,10	0.90	1.50	digin , desir	1.50	ga enq	1,60	
cateosin	0,0	0.70	1.90	2.30	40.40	2.20		1,60	1.10	GETT-repres
persin	0.0	0.50	0.90	0.0	483 -118-	0.60		0.30	व्यक्त प्राप्ति	engli date
tryosin	7-9 e-s	rup tea	ant set	WD 470	-			***		1.00
chymotryosin		36-40	ert est	**************************************	داده دینی	 pn	411		ratio WACT	0.60

In the process of hydrolysis of albumin in 24 hours here is liberated in the form of free amino nitrogen, the following percentages of the entire albuminous nitrogen: with catepain - 40%; under action of papain about 30%; by hydrolysis with trypsin only 15%; with persin 14%; and with chymotrypsin 9%. Denaturing of albumin by boiling has no effect on the extent of its hydrolysis with trypsin and chymotrypsin.

Chemical Composition of Albumin

We have already published certain data about the elementary composition of procollagen from the skins of rats and rabbits. In highly purified and multiprecipitated procollagen there is: carbon 49%, nitrogen 16%, hydrogen 7.5% (averages).

A small amount of phosphorus (0,15%) which is found in procellagen we are inclined to attribute to impurities which are difficult to isolate by methods available to us. On the basis of results from the ultraviolet spectrography we can already conclude that procollagen definitely or almost definitely does not contain tyrosine or tryptophane. On the basis of chemical reactions we have established that there is no tyrosine (Millon's reaction negative) and only traces of tryptophane (xanthoproteic reaction and voisene reaction). Reactions for sulphur are negative. We isolated from albumin histidine (in the form of nitroanilide), arginine (in the form of flaviamide) and lysine (in the form of nitroanilide). The general content of diamino acids about 5%.

6.

Procollagen contains over 10% of dicarboxylic acids (isolated by us in the form of barium salts). Albumin contains a considerable amount of proline and hydroxyproline. The reaction for carbohydrate is positive.

Conclusions

The albumin of skin which according to our tests possesses a high biochemical instability when isolated in crystalline form, differs in a series of characteristics from heretofore known albumins. It may be assumed that it is a forerunner of collagen and is, therefore, named procollagen by us.

It seems to us that there is basis to classify this albumin in a special group of connective tissue albumin and that procollagen will not be the only representative of this group. Procollagen is widely spread in the animal world and it is safe to say that there are not any species of vertebrates which do not contain this albumin. According to our tests, procollagen is present not only in the skin but also in a series of other tissues and organs of animals. In particular, it was possible to isolate from the tendons of a bull a crystalline albumin which resembled the procollagen of skin. It will be of interest to compare the identity of the procollagen isolated by us with the crystals of albumin in compounds of the sinews of a rat's tail as observed under a microscope by Nageotte and with the amorphous compound of collagen like albumin isolated from the skin.

We consider it our duty to express our thanks to our collaborator K. D. Leontgeva of the physiochemical laboratory for her great help in spectrography

Titles of Diagrams

- Diag. 1. Crystalline procollagen from the skin of a rabbit.
- Diag. 2. Crystalline procollagen from the hide of a calf.
- Diag. 3. (explained in text)
- Diag. 4. Ultraviolet absorption spectrum, 4% solution of pro-
- Diag. 5. Ultraviolet absorption spectrum, 6% solution of gelatin

7.

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V. N. Orekhovich, A. S. Konikova, K. D. Crekhovich
and N. N. Dobbert

CONCERNING THE METABOLIC TURNOVER RATES OF VARIOUS ORGAN AND TISSUE PROTEINS

Akademiia Nauk SSR Doklady, 71:105, 1950

It is well known that there is a continuous renewal of the parts of an organism; that its proteins, fats and carbohydrates do not remain unchanged after their synthesis and inclusion in organs or tissues, but are constantly being turned-over. Naturally, the study of this process is a very essential and important part of the deciphering of the functions of the organism. This is especially interesting from the standpoint of proteins.

For our studies we have made use of tagged atoms to determine the turnover rates of different organ and tissue proteins and their separate fractions. This problem has been studied by Schoenheimer and his associates (1), who made use of tagged amino acids. They only investigated the time for inclusion of these substances into different proteins, but not the overall turnover rate of the protein. Ussing and Krough? investigated the overall renewal of proteins with the aid of a heavy hydrogen isotope (deuterium), carried in the form of heavy water. They were only interested in the turnover rates of proteins of the skin, muscles and some internal organs.

We also made use of heavy water, studying the turnover rate of proteins from the organs and tissues of white rate, as well as some of their constituent fractions.

Heavy water was given to the animals at such a rate that after a few days its concentration in the body reached 1%. Then the animals were killed and proteins extracted from all organs and tissues. The proteins were carefully treated to remove physically bound water, dried to constant weight and incinerated. After being suitably cleaned, the water formed was analyzed for atom per cent deuterium by the flotation density method.

The data we obtained are presented in Table 1. It is seen that the fastest turnover rate occurs in the proteins of the liver, while that of the skin and muscles is the slowest. The rest of the investigated proteins lie between these limits.

We tried to characterize the turnover rate of proteins not only with respect to how fast they took up deuterium, but also as to how fast they lost it again. With this goal we worked with a group of rats which were given heavy water for twelve days and then not killed immediately, but after twelve more days had elapsed.

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Table 1
TURNOVER RATES OF PROTEINS OBTAINED FROM VARIOUS ORGANS

Organ	Excess of deuterium (at %) after intro- duction of D O	Turnover rate (%) of pro-teins	Excess of deuter- ium (at %) after stopping introduc- tion of D ₂ O	Differ- ence (%)
Liver	0.232	23.3	0.109	0.123
Intestine Spleen Kidneys Stomach Heart Lungs Brain	0.170 0.167 0.162 0.137 0.136 0.115	17.0 16.7 16.2 13.7 13.6 11.5	0.094 0.107 0.079 0.084 0.106 0.078 0.092	0.076 0.060 0.083 0.053 0.030 0.037 0.020

As we see from the data presented in Table 2, the rate of loss of deuterium from proteins of organs and tissues is not the same as their rate of uptake. For example, the uptake of deuterium by the kidneys ranks fifth, while it is second as far as rate of loss is concerned.

LOWERING OF THE AMOUNT OF DEUTERIUM IN PROTEINS OF VARIOUS ORGANS 12 DAYS AFTER IT CEASED BEING PRESENTED (%)

ORGAN	% DECREASE
Liver Kidneys Intestine Stomach Lungs Soleen	53 54 40 38 31 30
Heart Brain	29 19

There is the same lack of correspondence in the uptake and loss rates of the proteins from lungs and stomach. This is explained, apparently, by a difference in intensity of exchange rate of structural moieties between various organs. The disproportionally low figure for kidneys can be explained not only by the intensity of turnover of amino acids containing deuterium, but also, apparently, because of the dilution of their own structural proteins with those from organs which contain a low concentration of the isotope.

Together with the study of the intensity of the turnover rate of the proteins from different organs and tissues, we also study the turnover of various protein fractions derived from the same organ. We anticipated that by this method we would be able to answer the question, to what extApproved For Release 2003/67/29? ETA-RDP80-00926/3003400050022-5 f a given

organ is specific for individual fractions which go into the makeup of the organ. The figures we obtained are presented in Table 3.

Table 3

INTENSITY OF TURNOVER OF VARIOUS PROTEIN FRACTIONS
FROM DIFFERENT ORGANS

Protein	Excess of deuterium (at %)	Turnover rate (%)
Blood proteins Blood globulins Liver glovulins Skin globulins Albumins and globulins of the skin Collagen of the skin Procollagen of the skin Ossein (collagen ?) Muscle proteins Myogen	0.181 0.148 0.137 0.138 0.177 0.138 0.123 0.115 0.101 0.089	18.1 14.8 13.7 13.8 17.7 13.9 15.5 10.1

As can be seen from this data, the turnover rate of plobulins of the blood, liver and skin is practically the same, while the intensity of turnover of the collagen of the skin is much lower than the skin globulins. In this way we can see that the turnover rate of the fractions varies in each given organ.

Besides examining the turnover rates in normal adult rate, we also studied them in newborn animals and their mothers. For these experiments we fed a pregnant rat heavy water for twelve days before the birth of her litter. The same day that the rats were born they were killed, together with their mothers, and proteins from different organs analyzed for their content of deuterium. The results are given in Table 4.

Table 4

INTENSITY OF TURNOVER IN ATOM PERCENT OF DEUTERIUM

Tissue	Excess	atom-%	Protein tur	rnover 4
	Mother	<u>Newborn</u>	Nother	Newborn
Skin	0.070	0.254	7.0	52°6
Internal	0.187	0.254	18.7	
organs Carcass Muscles Head	0.070	0.264 0.252	7.0	36°5 56°7

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As we can see from Table 4, the turnover rate of the oroteins of the skin and muscles of the mother is markedly less than that of normal adult rats (Table 1). This indicates that there is a sharp decrease in turnover rate of muscle and skin proteins during organancy. The turnover rate for the internal organs of the mother is normal. The oroteins of all the tissues of the new born animals show almost the same excess of deuterium, which is much higher than any of the rates of the tissues of the mother. This may be explained, perhaps by the fact that all the organs and tissues of the embryo are being built up of free amino acids, without its making use of the formed proteins of its mother's organism.

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Presented: 6/31/40

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Tutanovskii A. A. "Concerning the proteins of skin."
Biokhimila, 12(4): 285-30. 1947

A crystalline protein resembling collagen is obtained from rabbit or rat skin by the following procedure: Shredded skin, freed from muscle, porous epidermis and superficial fat, is extracted with 0.1 M buffer composed of completely or partially neutralized organic acid (citric, succinic, oxalic, lactic, tartaric, adipic or glutamic). The optimum pH for extraction varies with the type of skin and the organic acid employed. In the extraction of rat skin the optimum pH is 3.5 for oxalic acid buffer and 4.1 for citric acid buffer, while in the extraction of rabbit skin with citric acid buffer the optimum pH is 5.7. Extraction with 5-6 ml. of buffer/g. of skin requires 12 hours under mechanical agitation. Extraction and subsequent operations are carried out at 7-8 C. The buffer extract is clarified by centrifugation and filtration. The filtrate is dialyzed against water, pH 7-8. When the buffer concentration falls to below 0.01 M, crystal separation occurs spontaneously. Crystals are predominantly needleartion, length 25-750 microns, diameter 5-6 microns. 100 g. of fresh rat skin yields 2.7 g. of dry crystallina protein. M. C. Brockmann

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A. L. Zaldes and S. L. Puoko (Akademila Nauk USSR Doklady, Vol. 73: 991, 1950)

ELECTRON-MICROSCOPIC INVESTIGATION OF THE EFFECTS OF ALKALIS AND PANCREATIN ON COLLAGEN

The electron microscopic examination of collagen has disclosed the presence a regularly repeating structural pattern in the fiber (1).

There is a strong interest in the effects of various treatments on this structure. Since the production of collagen by the skin involves the presence of specific fluids and enzymes of the pancreas—pancreatin, we decided to study the influence of these reagents first, especially since the illucidation of the action of lime on collagen is inadequate (2), and the literature on the enzymatic treatment is totally lacking.

Pieces of collagen, obtained from the frontal part of a steer skin, cut in a size 2 x 10 cm. were processed with a solution of lime-water containing calcium hydroxide in a concentration of lOg/l. They were kept in this suspension over a varying period of time, ranging from four days to two years. Before examination, the swollen fibers were neutralized in 5% bisulfite and carefully washed.

The enzymatic treatment of fibers kept in alkali for four days and then neutralized was carried out with a one per cent solution of pancreatin at a temperature of 37°C, for periods of 3 hours, 8 hours, 12 hours, and four days, with constant agitation all the while. With the protracted pancreatin treatment, we used a daily change of pancreatin solution.

Then the collagen was washed with water and dried in etherelcohol.

From the prepared tissues we cut sections in a freezing microtome; they were then dispersed for five minutes in a magnetostrictive apparatus with a frequency of about & kilohertz.

The dispersed collagen was placed on a collodion film and shadowed with chromium for increased contrast, using an angle of $15^{\rm Q}$.

The results show that treatment with limewater over a period of four days causes absolutely no change in the structure of the collagen (Fig. la). With collagen soaked for a month we begin to see disruption of the structure, which becomes more marked upon two months processing. Together with regions without local changes, we can see strongly deformed fibrils, mixed up among each other and with a loss of periodic structure (Fig. l,b).

A complete lack of regular structure is seen in collagen after two years of treatment with calcium hydroxide solution (Fig. 1,c).

We thus conclude that the limewater has to act a long time to cause complete breakdown of structure; its action is slow and irregular.

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A similarly unpredictable change in form is caused by the action of pancreatin. But the character of the alteration is quite different.

Enzymatic treatment carried out over a period of three hours does not change the typical electron-microscopic picture of collagen.

It is interesting to note that the indicated duration of treatment corresponds to collagen produced in a certain way, and that our experiments under these conditions showed absolutely no alteration in the finest details we could resolve.

In proportion to extended treatment in enzymatic solution, a destruction of the collagen starrs to appear gradually. In the e.m. oicture of material processed for 8 hours, we can see badly damaged regions together with ones that show clearly defined striations (Fig. 2,a). Treatment over the course of 12 hours causes partial unraveling and breakdown of the fibers (Fig. 2,b), while four-day digestion yields material with no regularity at all, giving just "debris" under the electron-microscope (Fig. 2,c).

In one of our papers (3), we noted that electron-micrographs of replicas of all collagen fibrils gave a definite orientation of perpendicular bands. Similarly with dispersed fibers, we noted this regularity of structure in all fibrils which could be observed close enough to each other. We can see in Fig. 3 that the dark and light regions correspond in adjacent fibers. This is an adequate proof of the organized spatial arrangements of the molecular micelles found in collagen.

We are indebted to A. M. Mikhailov and A. I. Frimer for their interest and work.

Central Scientific-examining Institute of the Leather-wear Industries of the USSR

Presented: 6/10/1950

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Translated by: Walter Stahl 4/18/1951

DIAGRAM CAPTIONS

- Fig. 1 Collagen treated with milk of lime, dispersed with sound and shadowed with chromium. Length of treatment: a-1 days, b-2 months, c-(beta)-2 years
- Fig. 2 Collagen after enzymatic treatment, dispersed with sound and shadowed with chromium. Length of enzymatic treatment: a-8 hours, b-12 hours, c-4 days
- Fig. 3 Ultrastructure of fibrile, dispersed with sound and shadowed with chromium

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Zaqdes, A. L. and S. L. Poxpko. The electron microscopic examination of collagen. Reports of the Acad. of Sci. of USSR LXV, 2, pp. 227-8, 1949. From the Central Research Inst. of the Leather-footwear Industries.

The electron microscope examination of the primary fibrous orotein of the skin, collagen, is of interest from two viewpoints. It is, of course, allowing the growth of ideas about collagen itself and its production by the skin. But it is also shedding light on the nature of fibrous proteins, in general, which are more widely distributed.

To the present time x-ray analysis has been used to examine the fine structure of collagen. This did not allow, however, the examination of the form and inter-relation of the elements of the subfibrils whose thickness ranges from 100-50 A.

For the electron microscopic examination of collagen we used both direct observation and replica technique. Our exmination by the latter method confirmed the earlier observations of the etructure of this material (1).

A more thorough method of direct observation was the result of sectioning in a microtome to a thickness of a few microns and then dispersing with ultrasonic vibrations of frequency of about 8,000 c.p.s. for 3-5 minutes. The preparations so obtained were deposited on films of collodion or aluminum oxide and shadowed with chromium to give more contrast (2).

Untreated fibers were examined, as well as those stained with various salts of heavy metals and extracts from wood of the oak.

All examinations of the fibers have revealed periodic bands which are arranged perpendicular to the long axis or as a spiral. The spiral form appeared very sharply in some of the photomicrographs.

When different forms of treatment are used, the width of the striations changes, revealing still finer bands within. Thus, material treated with tan-bark extracts shows an over-all spacing of about 700 A and a band width of about 170 A.

The polysterol-quartz replica of untreated collagen fully confirms the above findings (2).

Similar results are also obtained through the use of methyl-methacrylate-quartz replicas which avoid the necessity of heating the object during preparation.

Electron microscopic examinations of collagen fibers by American authors (3) gives the following picture of its structure. The wide bands of the collagen subfibrils consist of several smaller bands which show a characteristic, periodic density. The bands extend the width of the whole fiber. This conclusion is

not in agreement with our own experimental results.

If the subfibril really has the chain form given above, replicas of individual fibers should show the form of the fiber without its internal structure. Electron photomicrographs prepared by us, however, show a periodic structure consisting of two mutually perpendicular spirals of subfibrils.

These observations bring us to the conclusion that the molecular components of collagen show a contour structure.

The accompanying electron microphotographs (none) of untreated, shadowed collagen fibers (chromium) show the outer structure of the material. The shadows show the same periodic structure, proving the presence of the relief.

The observation of fibers at the moment of bursting due to the action of the electron beam on the fiber or supporting film has confirmed the above conclusion. The break always occurs at the lightest, and then the most thin, bands. The more dark, and therefore dense, bands then take on the form of an extended droplet. In the broken fibers the spacing between the light and dark bands is still more apparent, supporting the above hypothesis.

It is interesting to note that the broken fiber takes on a more uniform appearance when stained with phosphotungstic acid, apparently due to a more uniform absorption of the metal.

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Presented to Academy on January 15, 1949 by A. E. Oparin

Diagrams

Fig. 1. Unraveled collagen

Fig. 2a. Native collagen; 6-stained with phosphotungstic acid; -uranyl acetate stained; 7-oak extract treated.

Fig. 3A. Chromium shadowed collagen fiber; 5-polysterol quartz

replica of collagen
Fig. 4. PTA treated fiber distorted by bursting of underlying film.

Translated by Walter Stahl

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(Akademila Nauk USSR Doklady, Vol. 73:379, 1950)

THE FLEGTRON-MICROSCOPIC EXAMINATION OF FOLLAGEN USING THE REPLICA TECHNIQUE

The orimary advantage of the replica technique for examining collagen (1) lies in the fact that it does not require that the fibers be treated in any structurally-disturbing way (for instance, as in dispersion by sonic vibrations) and that one can obtain the impression of a block of material, according to which it is possible to judge the architecture of the tissues. The investigations of collagen by the collodion replica technique available in the literature (2) deal with samples of the fibers first dispersed by various methods (sonic vibration, colloid mill, mechanical unraveling with needles). By this method one obtains imprints of single fibers only, since by such treatment the histological structure of the collagen is destroyed.

Together with the study of dry objects by the use of polysterol-quartz replicas, we have worked out a method of getting replicas of wet materials. To achieve this goal we heat an aliquot of methyl methodrylate monomer in a reflex condenser, adding the polymerization catalyst. This is carried on until the liquid becomes slightly viscous. Then the top layer is cut off a piece of collagen in a freezing microtome and the remaining meterial is submerged in water.

After recovering the block from the vater, the surface chosen for examination is freed from excess water by using a piece of filter paper for drying (this process must be carried out rapidly). On a surface prepared in this way is poured the freshly prepared and slightly polymerized methyl methscrylate. In order to prevent drying, the specimen is covered with the liquid on all sides. The consequent polymerization process takes place on the object itself. The first layer of thin film is reinforced by a secondary coating of polymer until a thickness is built up which makes it convenient to remove the film (about 0.1 mm). Then the film is taken off and it is dusted with quartz in a vacuum to give a thin film (order of 100-300 A), applied on the side facing the collagen. The quartz film is freed from the methacrylate by dissolving the latter away with dichlorenthane.

Using the replica method we examined soecimens of collagen which were untreated and some which we prepared in various ways: in calcium hydroxide, enzymes (pancreatin), salts of heavy metals (PTA, PMA, and uranyl acetate), fixed in formaldehyde, primary sulfate of chromium and woody oak extract. In all cases we observed the presence of regular perpendicular bands of different densities (see figs. 1,a,b; 2,a,b; 3,a,b).

We were able to observe a greater differentiation on the replicas from wet fibers than or corresponding dried ones. Thus, in the case of the chrome-tanned material, besides the 634 A period there was one of 297 A. By ordinary observation in the electron microscope we observed roughly the same period (282 A) in collagen fibrils. In some of the replicas, obtained from the specimens treated with extract of the woody oak, we were able to see spherical particles lying on the fibrils (see fig. 4). In all probability, these particles are composed of tanning, not part of the collagen but only adsorbed on its surface.

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Vith an accuracy of about \$15 A. The periodicity is observed to the limits of each method of preparation (see table 1). We can divide the observations on the periods into three groups: 1—with a repeat period of 640 A in the wet state; 2—the period of dry, untanned objects (here too are specimens treated with PTA and formaldehyde) which ranges from 540 - 560 A; 3—a period for fibers treated with various tanning agents and dried, which is about 430 A.

Table 1

REPEAT PURIOD OF THE FIBER

Form of treatment	Repeat Period in A	Fiber Width in A
USING POLYSTEROL-QUARTZ	REPLICAS OF DRY FIBE	CRS
Untreated Calcium hydroxide (3 days) " " (1 month) Pancreatin (3 hours) " (12 hours) P.T.A. Formaldehyde P.M.A. Uranyl acetate Primary sulfate of chrome	535 576 546 546 586 409 409	650 980 820 1140 910 990 800 705
USING METHYL METACRYLATE-GOBJE	QUARTZ REPLICAS ON WI	et
Calcium hydroxide (3 days) Pancreatin (8 hours) Pancreatin (12 hours) Primary sulfate of chrome Extract of woody oak Formaldehyde	640 640 625 634 651 609	1200 1100 1100

Replicas have not only allowed us to illucidate the repeat periods of fibrils, but also the relative constancy of their width. For the various forms of collagen treatment the average width is about 0.1 microns. In the case of the wet fibers, tanned with the primary sulfate of chromium and formaldehyde, calculations show an increase in width of about 1020%.

Replicas of all the examined fibers show a definite orientation of the bands in different fibrils; the dark bands as well as the light coincide. This fact, and also the constant period of repetition, is characteristic of the collagen structure, indicating a definite soatial arrangement of molecular groups.

In conclusion, we are obligated to express our thanks to A. N. Mikhailov and A. I. Firmer for their interest and work.

Central Scientific-investigating Institute of the Leather-wear Industries of the USSR

Presented: 6/14/1950

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SHADE TURESBATURE (FAHRENBEIT) ALD MEASUREM HE OF BALWEALL FOR 1935 (Cont.d.)

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GHADE TOTTEBATURE (FARGUMENT) AND LABUR HAT OF BALLERALL WAR 1937

SYZIA

	:	July				ust		sj()	gru omb	Cal			ุยช <mark>ิ</mark> ย์	(.i*				OC1					
Date	. Max	∷ 1 n,	Rain Pall	: Mo	x, %1	n. Ruit Poll	1 :	Max.	Ein,	Rain Fall	:	Han.	ila.	Rein Tull	:	HOX.	in.	Smir 111		liax.	ain.	uir Pall:	
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2 :	52	37	~ ~~	:	57	37	47	:	49	35	0, 11	:	70	40	ag	:	72	69	**	:	90	<i>5</i> /1	-	:
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- () (· ·	53	54	0.15	:	57	2.7	==	•	55	36		;	66	56	co.	:	73	49	***	:	90	56	-	:
40 :	50	39		;	59	6	:	7	55	3 6	<i>5</i> -2	;	72	54	1979	:	72	45	No.	:	88	56	-	:
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<1 1 1.5 1	50 50	34	20	:	59 6h	36 113	0.10	5	6 5 63	35 40	ma.	1	91 91	55	-14	:	95 94	55 52	-	:	9 7 9 5	90 61	_	: .
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- K	50	36	u. 30	•	45	35	5		δã	1.1	20	3	ว์ดั	52			81		27	•	104	64	a	:
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	9.2	12		:					25	4.	-		.7	69	.•	•	بلان	53	€	:	95	59		:
w.	 .>+:	37	O. 11.	1	-		4		Ch	4.2	***	;	31	50	109		હેઇ	55	sub.	:	95	67	Wh.	:
2	54	36	w +		*	æ.				3.35			P-	-	-	2	37	54	-		-	6 22	-	
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STADS THE HOLD (SAIR MINISTE) AND MASURE OF STRIA

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STRIA

حيدجيود ۽ مستحق ع مي يد ۾ ۾		July			A	ugust				oterb				ctobe				ve.bc				censo		
Dote:	Kax.	Min.	Rain Fell		Max.	llin.	Rain Fell	:	Mux.	kin,	Rain 'all	:	Hux.	Min.	Reia Fall	:	Xax.).in,	Rain Yall	:	RX.	. in.	Rain Foll	:
1234967690123455789111112349676769012345578901222228550767695	1000589746677633663777335557773246832554	65666582218615566676665537 0 966557 0 96657 0 966557 0 966557 0 966557 0 966557 0 966557 0 966557 0 966557 0 966557 0 966557 0 966557 0 96657 0 96	0,04	40 48 75 87 48 88 87 48 80 50 58 88 88 88 57 78 78 86 78	106626223199988339579991297783843636	66766666666666666666666666666666666666	Fall	49	10473227950437905553836453344 8330477	554035423332522313398545 73357344 6667666666666665555355 555544		The second secon	344450 855556299901781124248897777780011111	5801865504555776044001119006565566554445555555555555555555555555	0.02	** ** ** ** ** ** ** ** ** ** ** ** **	82110999588123592221245656566666555555555666666666666666	6162447.05045144667188114450506666.0 5340454.44.333450556666.0 53404.55.4	0,02 0,17 0,50 0,25 0,05 0,05	: : : : : : : : : : : : : : : : : : : :	G0645104257717555307556651514917447	97877836658956660766652226 1297574 3334343354533353543355543554	0.10	
						the aut	**			no admirine.	~ . ,		Marine Calendari		~~~	**********							·	

SHADE TWIF BATURY (FAME USIT) & D SHUT III US VALUTAL OR 1938. (Jone 1d)

41. TABLE)

MONTHLY WEATHER REPORTS

HALFA - REFINERY SITE BY C.R.L. HAIFA REPORTED BY W.P. BELL

	1	9 4	2		9 4	5	1	9 4	44
	Тепрега	tures ^o r	Humidity	Tempera	tures ^o F	Humidity	Tempere	turesof	Humidi ty
	Dry Bulb	Wet Bulb	1 33	Dry Bulb	Wet Bulb	%	Pry Bulh	Wet Bulb	%
.ianuary	į			59	57.5	89	60	5€	79
February	62	54.5	63	60	58	87	6/2	55	65
March	65.5	56	58	6.1	59-5	92.5	67	60	68
April	73.	63.5	55	65	64	93	72	(66)	(73)
May	80	69	58	15	68	70	76	(70)	(73)
June	87	71,	53.	84	75	68	. 8≥	(73)	(72)
รับโต วันโต	59	74.	5×	85	76	68	86	(79)	(72)
August	87	74	55	88	78	64.5	82	73	66
Sepiambar Sepiambar	د	15.5	65	87.5	77	63			
October	84	/1,5	56	86	74-5	59			}
tedmesoli	79	69	55	71	68	63			l
ীনত প্ৰয়োচ ন্ত	57	53	80	70	614	72			

Remarkett (1) Figures in brackets are probably too high.

(a) Omitted figures were reconsignatered at the time
(3) This table comitted in durantified Data on Fari
Absolute or Artrone Values not given.

. 2 ..

Year	Press. @ noon			Tsn	арз	r a	. 1.	u r e	ob			Relat	ive Hu	idi ty	Ra	i n r e	1)
dt.	Mean	Averag:	Dry B	ilb Temp.	Extre	me Dry	Bulb	Temp.	Average	Wet Bu	ılb Təmp.	Ì	%			Highest in	24 hours
K onth	(unches)	Day	Night		Kax.	Date	Min.	Date	Day	Naght	Mean	Day	Night	Mean	Motal	Inches	Date
Jun-	29,77 29,76 29,76 29,85 30,06 30,05 29,96 30,04 30,03 29,96 29,78 39,84	86 87,5 86 80 76,5 63 70 74 80 84 87	77 78 77, 5 69, 5 64 59 54 55 69 777	815 83 82 75 70 61 565 59 64 685 745 785 825	97 96 94.5 98.5 77 76 81 93 99 98 99	15:8 10:9 24:10 18:11 7:12 27:1 28:2 27:3 12:4 4:5 28:6	54	29:7 9:8 30:9 31:10 14:11 21:12 14:1 10:2 16:3 17:4 15:5 22:6 21:7	77 78 75.5 68.5 66.5 56.5 54 57 62 63.5 69 72	73 74 72.5 63.5 59.5 53.5 51.5 53 55 58 68 73	75 76 74 66 62.5 55 55 55 58.5 61 66 70 74.5	66 66 62 57 57 68 72 70 66 58 58 57	85 83 79 73 78 70 85 90 85 75 72 77	75 74 70 67 69 79 80 75 66 65 72	NIL NIL NIL 0.396 trace 4.335 10.556 1.405 0.460 0.603 0.210 NIL	0.190 trace 1.342 2.890 0.465 0.232 0.256 0.116	14:10 29-11 14:12 13:1 17:2 5:3 16:4 12:5

Remarks: Omnitted figures were not registered at the time.

WPB/HNA.

TABLE III

PALESTINE METEOROLOGICAL SERVICE.

MONTHLY WEATHER REPORTS

HAIFA MT. CARMEL

SUPPLIED BY PAIRSTINE GOVERNMENT

	TEMPERATURE Near Pressure TEMPERATURE			SUPLLI	D.L.	MIL N	37 771.	, GO11.	40,000,00									
Y-ar	, ,			те	M P	E R	A T	U R	E G	F		Relat	ive (%	Humi	dity	Rs	ın fa	1 1
&	Mean	G .	М,	T	М	EAN		E	X T R	E M	E	G ,	Ħ	T	;	Total	Highes! in 24	Hours .
Month	Inches	0 6	12	18	Max	Min.	Mean	Max.	Pa+s	Min.	Dat=	06	12	3.8	Mean	Inobes	Inches	Date
1946 December	29,94	55	63	58	64	52	58	72	2:12	46	22:12	69	59	69	66	4.095	1.350	14:12
1947 January	29.90	52	56	54	59	50	54.5	68	27:1	44.5	14:1	77	74	75	55	11.949	3.185	13:1
February	29,96	55	61	56	63	50	56.5	75-5	28:2	43	4:2	69	68	76	7.1	1.145	n. 362	17:2
March	29.97	60,5	66	60	69	5 6	62.5	87	27:3	50.5	16:3	69	60	75	68	0.386	0.200	17:3
April	29,96	65	69	63	72	60	6 6	92	11:4	49	17:4	62	60	69	64	0.524	0.240	16:4
May	29.86	71.5	75	68.5	79	64.5	72.5	93	4:5	59	13:5	63	51	72	63	0,240	0.086	9:5
-	į	73.5	17	72	79-5	-68	-/4	84	28:6	64	2:6	74	69	85	76	-	_	-
]		

x) Corrected to sea level.

60.

(S) Preserve algoris and business of the party

5a.

61.

recipi tation

Moar Armual Rainfall

	Height	Inches	i endod	Authority
Station	ar det			U.C. BYBON TO SA
) Karan aran aran aran baran karan karan aran karan karan karan aran	(approx.)	TOPINAL	rate inconstruction and construction of	many diph conjugation of the control of the second of the conjugation
* p. % &				
Loner Belta			2076 0	Companie
Pap _	7	6,6	2936-9	iort
Magil	7	5.2	1926-39	'hailway G
Shualba	60	5.7	1925-39	นึกว่าม ^{ีร} ิง
Ghubai shiye	13	3.2	1,928-39	Railweys
lir .	2.3	2.9	£#	ęę.
Amera	3Ô	8.3	1936 -9	osto and Telegrapha
Faddatas: 1: 5	24		,	
Upper Dolta				
Jamewa	୧୦	3.0	2928-39	Cailways
	70	4.9	#	2. L.F.
Dimmlye			18	Rellways
Hilla_	<u>90</u>	5.9	£†	un i radi o
Karbala	35	2.4.	•	
Kindiya	175	3.8	? *	Irrigation
Habbani.ya	$M_i L_i$	5.8	19 37~9	R. A.P.
Cala Sikar	43	52	2936 -9	fosts and Tolographs
Eut ol Impra	52	5.9	ti	19
Hinaidi (Baghded)	12.0	5.5	1928-3 9	· Rada Pa
	22.3	3.5	71	Roilways
Samarra			2026.5	Absta and Telegraphs
Pendali	350	hiest.	1936-9	noses and taxolingum
Jasira and Suphrates above Roredi				
Baiji	459	3.5	3,538,09	Infa!!
	1, 950	19.5	1956-9	Poste and Telegraphs
Sinjer	2,200	17.2	3956-8	lolies
Bir Ugla				
Esditha	450	5.6	1934-9	L. E. C.
Ana	500	5.4	1936-9	osts and Telegraph:
Assyrian Plains and Foothille				
rable st. (Mansur)	220	6.7	3.92∂ ~39	Bailmays
Qaraghen (Jalopla)	390	7.3	19	79
Thanagin	660	12.0	295200	41
Tus Khurmatli	720	8.3	3.928-39	15.
Iftikhar	670	8.6	19	ષ
	1,00 0	14.8	17	I.P.C. and Railways
Kirkuk			1936-9	I.t.G.
Ditis	780	17.2		
Erbil	1,360	38.2	1936-8	kosts and Telegraph
Nosul.	730	13.0	3,925-39	Roberto
Transfer at Standard and the se				
Kurdish Mountains	9 200	. 12 2	3935-9	tosts and Telegraph
Helabja	2,300	15.3 20.3	- 220°Y	ుండులు ముట్టు మాడియల్లోనిక్లు ఆ
Sulabenniya	2,750	32.9		
Diena (Ruwandis)	0 ₉ 700	hJ.A	1936-7,	
			2539	
Acra	2,500	36, 9	1936-9	e
Amodia	3,500	43.2	13	19
Zakho	5,450	10.0	-4	Ħ
Western and Southern Beses a	- 40 Tem -	- · ·		
	3,040	6.6	2034-9	I.P.C.
71	المتعادية مريع و		يق المساهل كي الرياطة او	201.000
<u>10.</u>	2,340	5.9	:	ता. वर्ग
112	£,930	5•I	.; pr	., G
H3	2, 550	<i>1</i> , 8	ta.	14
75 23s	2,08 0	5.8	1929 -39	R.A.F.
iu iba				
Sukhaib	3,000	3.7	479049	k'olige
Sukhaib	3,000	3:7 	3036-9	Folice
	3,000		್ರಾಂಕ್ಷ್ 26A0054000	

Precipitation Reinfall in Inches and Number of Rain-days

	Jan.	Feb.	Mar.	Apr.	May	June	July	Aug.	sept.	Oct.	Nov.	Dec.	Year	Max. in 24hra.	
SHUAIBA (13-14 year	do 'en	servat	ions)		0.3	0.0	0.0	0.0	0.0	0.1	1.1	1.0	5.7	2.2	
Inches	1.3	1.2	0.4		0.1	0.0	0.0	0.0	0.0	0.8	4.0	3.7	21.3	witte	
Days	6.5	0.5	2.9	2,4	0.5	0.0	0,0	0.0	3.4	• • •	, -				
DIWAMIYA (10-12 ye:	ars' o	bserva	tions)								0.6	0.9	4.9	4.5 €	
Inches	1.0	0.9	0.2	0.5	0.7	0.0	0.0	0.0	0.0	0.0			4.7	40,5	
Days	• •	• •	• •	• •	1/2	ot land	W70.	• •	••	• •	••	• •	••	••	
HINAIDI (13-14 yea:	res [†] ob	ae rvat	iona)								- 4	3.0	a a	1.5	
Inches	1.2	1.1	0.3	0.4	0.4	0.0		0.0	0.0	0.1	1.0	1.0 4.5	5.5 28.3		la.
Days	4.9	5.5	0.3 2.6	2.7	1.8	0.2	0.0	0.0	0.1	1.5	4.5	4.5	20.0		T TOWN
RUTEA (9-10 years'	റ്റുണ	watior	ıs)								0.7	0.7	3.8	1.8	1
Inches	0.9	0.7	0.1	0.4	0.2	0.0		0.0	0.0	0.1	0.7	0.7 3.5	20.5		
Days	4.3	3.0	1.0	3.0	1.7	0.0	0.0	0.0	0.3	1.0	3.7	2.9	20.5		
MOSUL (13-14 years	1 ohme	montic	mal										17.0	72 Je 🔏	
	9.7	3.0	1.6	2.8	0.5	0.0	0.0	0.0	0.0	0.2	1.8	2,0	13.0		
Inches Days	8.8		6.8	7.4				0.0	0.1	1.6	7.8	9.1	59.9	-	

In the rainfall, the figure 0.0 indicates a mean rainfall of less than 0.05; in the rain-days the figure 0.0 indicates a total of less than 5 days in the month over a period of 100 years, i.e. less than 1 rain-day in 20 years.

Abnormal thunderstorm in May.

Abnormal downpour in November.

5b.
SYRIA & LUBANON.
Precipitation
Rainfall (inches)

	Yrs. V	Jan.	Feb.	Mar.	Apr.	. aw	June	July	Aug.	Sept.	Oct.	NOV.	Dec.	Total
Approb Aproxima includence imperior training distributions.	COSIES	Jame	2600	110140	ADI's		0 4434	G Vessy	11000					
Coast														
Alexandretta	10-11	3.0	3.5	2.5	2.2	1.8	1.5	0.5	0.6	2 "0	2.9	2.9	3.9	27.3
Beirut	61	7.3	6.4	3.5	2.2	0.6	0.1	0.0	0.0	0.2	1.9	5.1	7.5	34.8
Haifa	14	7.1	5.7	0.9	0.7	0.1	0.0	0.0	0.0	0.0	0.5	2.7	6.7	24.4
Countains														
El Kar eya	10	11.1	13.7	7.7	3.7	1.4	0.3	0.0	0.0	0,3	2.0	6.5	10.0	56.7
Depression														17 6
Flome	6-10	2.9	3-3	1.0	1.0	0.2	0.0	0.0	0.0	0.04	1.0	1.8	2.0	13.4
Ksara	8-11	5.6	6.0	2.3	2.5	0.3	0.0	0.1	0.0	0.1	1.1	2.2	4.5	24.7
Steppe and D	esert											٠,		2 c N
Aleggo	5-7	3.0	S-8	1.0	1.3	0.4	0.1	0.0	0.1	0.0	8.0	2.4	3.2	i) ol
Selemiyeh	2?	2.4	3.5	1.6	1.2	0.8	0.2	0.0	0.0	0.0	0.4	0.5	2.2	12.8
Damascus	7-30	1.7	2.1	0.4	0.5	0.2	0.0	0.0	0.0	0.7	0.4	1.6	1.6 1.1	9.2
Palmyra	6-9	1.0	0.8	0.2	0.5	0.3	0.0	0.0	0.0	0.0	0.3	0.3		4.5 6.3
Deir ez Zor	5-10	1.6	1.0	0.3	0.6	0.1	0.0	0.0	0.0	0.0	0.2	1.5	1.0 2.9	16.1
Urfe	10	2.6	3.0	2.6	1.3	1.4	0.0	0.0	0.0	0.0	0.4	1.9	2.9	4004
				-			Andrea bereigne begen aus der							Annahum of the Annahum of

SYRIA & LEBAROR .

tenne e e i maso d'izraci	obens.	€ C.32 ·	វាទ្ធមក	mar.	Apr.	бау	June	July		Sept.	Oct.	liov .	Des.	Total
্নান্ত্র ১৯৯৬ ১৯ সংগ্রহণ শুভ্রাইঞ্চ ১৯ ব শুনুদ ১৯৮১ বিজ	9 61 34	9 35 14	12. 14. 13	8 11 5	7 6 4	5	<i>t:</i> 1, 0	2 1 0	5	5 2 0	6 4 2	8 9.6	9 23 11	76 78 56
1940 pp 1947 St. 100	10	15	12	31/1	7	5	2	0,2	0	ì	ϵ	3.0	13	54
uproraism Load Salva	2-7 8-17	9 24	11 12	9	56	0.7	0.1 0.7	0 0	0 0.1	0.1 0.9	2 5	lu 8	9 11	50 68
EDINGO ARA DE TEST	0ert 27 7-8 5-8 5-8	11 12 6 6 8	11 15 4 5	7823331	453539	3 0.7 0.9	0:5 0:1 0:3 0:4 1	0 0 0.0 0.0	0.2	0 0 2 0 0	30 2 2 PM 60 PM	745448	11 10 56 59	50 60 33 31 29 65

m Rain-day is one with 0,008 inch rainfall or more.

Number of days with Enil

er er sere er sekkreig e	7859	******************	MALE PARKET AND AN AND ARMS		mane data secuent species	etudosta de Cale	e derenderen der den zweid	10. / 10. / 10. / 10. / 10.	A 2001-04	THE COMPANY OF THE PROPERTY OF	and an extension of the contract of the contra	paginga	···wittedare/P1。 4 / THE	and the state of t
	oens.	Jean		Mara	A. I.	. Lay_	नुमाहः	<u>July</u>	1410	ا علاقات	Deta	Roy .	೨೮೦	Potel
* - \$? 5		1. 4	7 6	$a : \mathbb{R}^n$	1.0	0,2	0.0	O O	α, α	0.1	- 14	₹.₽	€.?
				;										
4- 1	4	€.6	0.6	5.6	0 6	C/C	0.5	0.0	0.0	1 - 1	₩.2	9.6	ينا. خ	7 0

Sc. 17 - The President (in inclus)

	Jan,	Feb.	i .ux.	Apr.	nay	Juno	July	Aug.	opt.	Oct.	Nov.	Deo.	Year
l. Coast Acre(1928-38)	-												
ligan	5.5	7.9	0.9	0.7	0.2	trace	0.0	0.0	0,2	1.3	4.1	3. 7	24.5
Hax.in 24 hrs. Baifa(1921-34)	1,2	1.1	0.5	0.4	0.1	trace	0.0	0.0	0,1	0.3	0.9	1.4	
Mean	7.1	5.7	0.9	0.7	0.1	0.0	0.0	0.0	0.0	0.5	2.7	6.7	24.4
Max. in 24 hrs.	3.4	2,2	1.0	1.0	0.4	0.2	0.3	0.1	0.1	1.9	2.6	7.2	
Max. in month	12,1	10.5	6.0	3.5	2.1	0,6	0.0	< 0.1	0.5	3.3	13.6	13.9	39. 5
and year Elin.in month	0.7	0,2	0,9	<0.1	0.0	0.0	0.0	0.0	0.0	0.0	< 0.1	0.0	16.8
and year				(0 0					•				
Jaffa(1902-12)	g., p.,	~ ^	A ***	0.0	0.3	/A 3	0.0	0.0	0.2	1.4	J. 2	ى ق	23.6
Mean Max. in month	5.6 8.6	3.9 5.9	2.7 7.6	0.9	0.1 0.6	<0.1 0.2	0.0	0.0	1.1	3.9	5.5	12.5	28.4
and year	0,0	202	6.00		•••	•,	_					-	-
Kin. in month	4.0	1.0	<0.1	<0.1	0.0	0.0	0.0	0.0	0,0	0,0	0.4	1.9	17.3
and year Haza(1921-34)													
lean	3. Ü	3.4	0.7	0.7	0.1	0.0	0.0	0.0	0.0	ت.o	1.4	3.6	13.8
lax. in 24 hrs.	2.7	1.8	0,8	2.7	0.4	0.1	0.0	0,0	0,0	ز.1	2.1	2.9	-
2. Inland													
Jenin(1921-34)									2.0	A =	4 57	2. 9	20.0
Mean Max.in 2h hre.	5.1 2.2	2.4	1.1	2.5	0.1	0.0 0.1	0.0	0.0 0.1	0.0	0.3	1.7 2.0	4.1	19.0
Aux. In 24 hrs.	6,6	2.0	4.4	د. ۶	0.5	Usa	0,13	V2	0.0	مكر و مكر	~. 0	Ciri go ida	
Mean	6.3	4.6	3.7	1.0	0,2	0.0	0.0	0,0	0,, 0	6,0	3.4	7.1	27.1
Dax, in month	14.2	10.3	5.9	2.8	1:0	0.0	< 0.1	0.0	$\langle 0, 1 \rangle$	2,5	9.1	12.8	37. 9
and year In. 11 month	1.9	3.7	< 0.3	0.0	9.0	3.3	0.0	2.0	0.0	0.0	0, 5	0.7	3.8.5
and year	Sea 2	s8s ; ∮	< ¥ 0 € €	ters at	120 12	v. J	101 M	Jan	4,0	100 W	رون	- 4 5	2.000

PALASTRE
Procipitation (in inches) (Cont'd)

CO-CONTRACTOR CONTRACTOR CONTRACTOR AND ANALYSIS AND ANAL	Jan.	Feb.	lar.	Apr.	Lay	June	July	્રેપાટ.	Sept.	Oct.	Nov.	Dec.	Year
Beit Jemal(1930-8)				THE STREET SAME THE COMPANY OF THE C	immainmentusting upo vill	THE THE WEST STEEL COME STATE AND ADDRESS.	destror din rederator za serio spelje	digipanumini (1904)dibunusikudiki <u>da</u>	Entreller, ter der ollerentpringsså			and the second second	
Mean	4.3	4.0	1.3	0.7	0.1	0.0	0.0	0.0	trace	0.6	2.8	2.5	16.1
Lax. in 24 hrs.	1.2	1.5	0.0	0.5	0.1	0.0	0.0	0.0	trace	0.3	1.4	ō.8	2032
Jerusalem(1918-34) Hean	4.1	5.3	1.3	1.0	n 3	0.0		0.0	0.0		• •		
liax. in 2/1 hrs.	30	3.4	1.1	1.5	0,1	0.0	0.0	0.0	0.0	0.2	1.2	2.9	15.9
flax in month	3.9 14.5	12.6	12.	8,5	1.3	0.2	0.0	trace 0.1	0.4	0.9 2.3	2.2 8.0	3.0	1.7 6
and year			acat year () to	Street Sales	J. 40 J	335	0.0	0,1	0.0	400	0.0	16.5	41.6
Min. in month	0.1	0.2	0.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	<0.1	0.5	13.3
and year										- • •		-0,,,	>->
1 Latrun(1901-12) Mean	6.5	2 7	n r	*1 ***		/ - =							
Max. in 2h hrs.	6.5	3.3 2.1	2.8 1.6	1.3	0.3	40.1	0.0	0.0	<0.1	0.9	2.4	5.2	22.8
Max, in month	4, 6 8, 9	7.7	6.5	1.9	0.5	0.0 0.2	0.0	0.0	0.3	1.4	1.6	3.1	20.5
and year	0,5	101	0.5	4,0		000	0.0	0.0	0,1	4.5	7.0	12.5	32.5
kin.in month	3.4	0.0	0.6	0,2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.9	15.8
and year			•		•	- 5 -			•••	Og C	0,0	20)	100
Mebron(1896-1914)													
Hean Hax. in month	. 16,2	4.6	3.4 8.6	ુ. 0	0.3	<0.1	0.0	0.0	< 0.1	0.5	2.1	5.1	24.3
and year	13.9	12.4	a, €	8.3	2.2	1.3	0.0	0.0	0,2	2.5	6.1	14.1	39.8
Min. in month	1.6	0, 2	0.2	<0.2	0.0	0.0	00	0.0	0.0	0.0	0.0	, ,	n / n
and year		-1· III	~ 0 14	1000	1.7 4 1.7	U o U	₩,,0	0,0	0.0	0.0	0.0	1.1	10.3
eersheba(1921-兆)													
Mean	1,9	2.8	0.7	0.4	0.1	0.0	0.0	0.0	0.0	0.1	0.8	1.0	7.0
Max, in 24 hrs.	2,0	2.	1,(1.0	1.0	0.0	0.0	0.0	trace	0.4	1.4	2,5	140
										m 3 mb	0 - }	ر ه	

Price life Procepitation (in inches) (Contid)

researched that the consideration is a second as the constitution of the constitution	Jan.	Peb.	llar.	Apr.	Lay	June	July	Aug.	Sept.	Oct.	Nov.	Dec.	Year
3. Jordan Valley													
Tiberias(1890-1907)	i. ==	7.0	2,€	1.1	0.2	0,0	0,0	0.0	(0.1	0.6	2.9	5.1	20,2
# Fax. in 24 hrs.	4.7 2.7	3.2 1.2	2,0	1.5	0.4	0.0	0,0	0.0	₹0.1	0.5	1.6	2.4 8.8	٠
Max. in month	11.2	6.5	5.6	3.0	ŏ. Š	0.0	0.0	0.0	ζ0.1	2.1	6.7	8,8	27.7
and year				_						0.0	0.0	1.7	14.4
lin, in month	0.5	<0.1	0.1	0°0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.1	T424
and year													
Beisan(1930-8) Mean	3.1	2.8	0.7	0.4	0.2	0.0	0.0	0.0	0.0	0.7	1.2	1.4	10.5
Max.in 24 hrs.	ő. 9	0.9	0.1.	0.3	0.1	0.0	0.0	0.0	trace	0.5	0.4	0.6	~
Jericho(1921-34)				0.1		2.2	0.0	0.0	0.0	0.1	0.4	1.2	5.0
Lean	1.3	1.3	0.4	0.4	0.1	$0.5 \\ 0.0$	0.0 0.1	0.0	trace	0.2	1.0	î.5), v
hax, in 24 hrs.	1.0	1.3	0.6	0.7	1.0	0, 5	00.0	0,0	01.000	0,12	-, -		
Dead Sea, North end (1954-7	0.6	0.4	< 0.3	<0,1	<0.1	0.0	0.0	0.0	0,0	0.4	(0.1	0.4	1.9
Dead Sea, outh end(1935-6		- ,	•					c. c.	2.0	0.7	0.7	0.4	1.2
lean	0.3	0.1	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.3	0.3	U _a (‡	شه و شد
1 Parameter of the Africa													
4. Transjorden Amman(1924-41)													
// loan	2,5	3.5	0,8	0.6	J.1	0,0	0,5	0.0	0.0	0.3	ڌِ ۽ تِ	1	10.5
Lex, in 24 hvs.	2.0	2., 3	1.1	1. (9.9	0,0	0.0	0.0	0,0	Jail	3.1	2.1	reg
FAMILIANDE SECTION SANDALL TO MARKET AND THE SECTION OF SHOULD AND ARRESTS.	A CO TO A SECURITION	encales constant and	Same according to the same	rges or discharge s	1 71, 3	20 m. ma mameria	unarram sitti ir etin 21 tae	an or address of the Control	Wat Water Land Adv. District				

0301-1000 # 0301-11 # 0300-14 # 0300-9 # 0300-9 # 0300-9 # 0300-9 (6) GROUND THE PERAPTURES AT ALTHUD, INA.)

SUPPLIED BY MIDDLE EAST PIPELINES, LONDON - D. BUCKHAM.

FARIOD DECEMBER 1946
TO HOVELDER 1947.

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

MALCO REFUGED CROUND THE HEATURE AT A DEFIN OF A WEST.

	The state of the s													
	Gr.	0200	04/00	0600	g800	10 00	1200	7 ,4. 00	1600	1800	S(A	22.00	Brs.	
سد. روز.	,	76,8	76.7	76.7	77-9	78.1	78.6	78.8	78-3	76-3	7707	77.5	O.	
. •	ا بغر													
	397	25	93	93	96	97	97	96	96	95) [为	לפ	بلأو	
	r Bakan menengga						-	LINEAR PERSONNELS	Standschaft in etablic	i Complete (Accordante (1980))	Lucian	Samuel of Marketon (* 1776)	:	

Yearly Characteristics.

Average 77.7°F. Nininum 55 °F. Neximum 97 °F.

The object of 50°F for the minimum temperature to not considered contents and the constant and the constant

ALWAND REPITERY SHADE TE FERATURES.

7 x 06	i i de se
the state of the s	
10 30 69.2 66.7 64.4 63.7 70.6 78.6 83.4 85.2 84.6 81.6 73 173.5	l on
75 34,2 32.5 31.5 50.5 39 44 45 47 47 47	i ·
Table 102 100 99 98 102 109 117 117 114 134	;

Monthly Characteristics.

Average 74.9 Minimum 30.5 Maximum 117

Charlied by hipping last pipalines to account to account

GRAPIS, CAL BESULTS

desperature as a hourly intervals.

STAND COTTONY GROUD THE EPATHER AS A DESIGN OF BURGET

ine	oc.	5200	ovvoc	⊅600	0800	1.000	1200	3400	7600	i6cc l	iidac	. <u>2</u> 20.	las.
vge.	63.7	65.3	63.3	€3.3	64.1	61 ₁ 9	65.4	65.3	40% 14	Q. i	Cate .	I, -, +,	ь , баў 1
v o	100	>5	55	55	5 7	58	59	50	<u></u>	jë:	. 5	. 74	. 2
.) د ر و ا	69	69	ú 9	69	70	71.	72	71	71.	7u	Æ	70	÷F.

isaly Temperature Ronge:

Marting Characterist with any

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WEAVED REFERENCE - SHADE TWO TER-STORE

ীয়ধ	•		1	1 1			i		1	2000			T #9
Ago.	61.1	1	1	i .									G _p
la.	51	50	49	47	51	56	<i>5</i> 9	33	ا مراس	3,1	3 7	. Die	ŭ ".
#X.,	76	74.	74	72	80	84	86	57	ing		•77.	π,	O.F.

Monthly Characteristics.

Average 64.9°F Minimum 47 °F Maximum 89 °F

71.

CCTOBER 1947

A .- LO OIL RESULTS

femmerature at 2 Hourly Intervals.

ALTHOUGH OF A TWATE IN PRAINING AT A DEPTH OF A PERM.

20	.)n.	0200	المناسبة	- 16.0 	rever	2,000	1290	1400	1600	18 00	2000	24.00	
ု့ပ	75.5	73.3	1737	"" t, G	732 j	75.2	753	76.0	75.9	74.5	Pac	12-1	. J
73	69	[_E ,€,	1 ->C	tat.	76.	190	7 0	70	70	70	<i>1</i> .j	Áÿ .	· - 48
$\mathcal{L}_{\mathcal{O}}$	76	78	. 47	2+	80	80	80	80	30	76	7 5	79	7
	-		100		;	7 704				أحديد عديدي			

Dauly Compete the proper

Avorage 5 1200 ... Lineaue e Lineaue 5 1, Monthly Characteristics

Average 7%-307
Dinume 68.448 (Odder ors
do:Lana 80 48 (Odder ors
Look)

The state of the s

	1			• • •	•			ten er e v	ي. ، ،	* CERTS POLICE CONTROL	fer fere attitude annotation to the state of		
·.•	: 00, 	0200	1416 J	i ti	GÖL U		ikuti	بالنهد	7900	3800	2000	22 . 45	Pio n
				,		7 1 1 mm					i .		
	\ \ \ \? . \ ? . \ \ \ \ \ \ \ \ \ \ \ \			(A) 1	. · · · · · · · · · · · · · · · · · · ·	23. 3	57.2	20-3	104.6	M _s , s	179 51	72 2	
•	כר	5 i	•								j Got		54
		W.	•	(#)	J".	i in the second	379;	199	, e. va	: 52		300	
٠	indexis year in the			***						r :	t were a visite a		

hally Japacteristics.

797166 1803 10060 77 (0600 55 5 06 00 5 3 100 50 7 (0600 55 5 06 00 5 3

72.

SEPTEMBER 1947

STATISTICAL RESULTS

Temperature et 2 Hourly Intervals.

ALWALD REFLIERY GROUND THE BRATURE AT A DEPTH OF 4 FRET.

line	00.	0200	04.00	0600	0800	1000	1200	14 00	1600	18 c o	2000	2 2, 00	Ero:
raße°	83.8	85.7	83.4	83.1	85. ₄	86,2	86.4	86.7	86.7	85.7	84.5	84.2	ှာ ရှ
in	77	77	77	76	78	76	79	80	80	78	78	77	ं कु
iar.	88	88	88	87	90	90	90	91	93.	90	9 0	88	O.B

bily Temperature Range.

Monthly Characteristics.

Average 3,60°F Jazimum 5

Average 34, 90P Minimum 76 °F (0600 hrs 23 Sept.)
Maximum 91 °F (1400 hrs 1 Sept.)

ALMAND REFLIERY SHADE TELFERATURE

ire	00.	0200	0400	0600	0800	3000	1200	1460	1600	1300	2000	22,00	lírs.
nge.	79.0	76.1	74 ₀ 2	72.0	82.3	91.8	97.2	99.2	9E. 7	94-9	8. 38	83.8	3·gr
iin.	74	72	68	64	/5	87	90	92	23	86	80	76	alb
ax.	95	93	90	86	93.	3.06	209	123	320	104	98	96	Olle

Monthly Characteristics.

19/31 AUGUST 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWARD REFINERY GROUND THE PERATURE AT A DEPTH OF & PART.

12.390	60.	050 C	0420	0600	08 00	1000	1200	22,00	1600	1800	2000	22.00	Arc.
*ge.	90.5	90.1	90 .0	90.0	93.3	93.6	94 ₆₀ 0	94.3	95.9	93.2	92.3	912	eoft.
in.	88	88	87	87	90	90	90	91	97.	30	89	89	K* Pr
Ax.	93	93	93	93	95	95	95	95	95	4 5	95	94	o _k

Daily Temperature Range,

Konthly Characteristics.

Average 4.309 Minimum 2 of Maximum 7 of Average 92.2008
Minimum 87 98 (06.30 km 30 Ang.)
Marinum 95 98 (06.00 km 19 Ang.)

ALWAYD REPT MAY SHADE TELLERATURE

Ž ipi	co.	0200	0400	0600	0800	1000	1200	1400	į	1800	ŧ.	22.0 0	4
Fge.	86,8	83.2	80.1	77.7	91.4	9 9 .8	104.9		106.5	ţ		93.3	0 %
in.	80	76 .	72	70	82	92	301	103	103	100	92	87	O.f
AX.	96	94	93	93	101	108	112	114	113	717	104	99	Op

Monthly Characteristics.

Average %, 40% Minimum 70 °F (0600 hrs 31 Augs) Maximum 114 °F (1400 hrs 26 Augs)

JULY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALMAND REPTINERY GROUND TREFFERATURE AT A DEPTH OF A FERT.

2.25%	00,	0200	orroo	0600	0800	2.000	1200	1400	16 00	1800	2000	22.00	Bra.
\$9a.	91.0	50.6	90.5	90.8	93.7	94-0	94.0	94.1	94-3	94.0	92.6	92.9	Oge
an i	85 ·	.15	36	87	92	92	92	92	92	92	30	90	og
out.	ac.	95	93	93	%	97	97	96	96	9 5	95	95	Ongo

19 Dy Temperature Range.

Monthly Characteristics.

Average 3.56°F Tireom 3 °F aximus 5 °F

Average 32.7°F Minimum 85 °F (0000 has 51 July) Meximum 97 °F (1000 has 51 July)

ALMAND REFIGERY SHADE TENYATURE

4.195	00.	* 6 ? 00	0400	3600	9800	1000	1200	1400	1600	1806	2009	22. 00	i Lara,
je	92.4	87.7	82.1	83.6	95 .0	102.6	108.9	110.6	110.4	لادة.5 5.06	100.4	97.a.	i ug
d-	85	80	73	77	9 0	96	104	105	105	LOL	55	<u> </u>	i Ch
: t .	102	100	99	98	102	109	127	117	117	طرن	107	104	الإن

Monthly Characteristics.

Average 98.80p Minimum 73 °F (0400 ice 4 July) Haximum 117 °F (1200 ice 20 July)

JUNE 7.947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REFINERT GROUND THE PERATURE AT A DEN'TH OF 4 FIRET.

Ĵwat	20.	0200	0400	0 600	0800	1000	1200	1400	1600		2000	22,00	Hrv.
sRe"	85.7	85.3	85 。0	85.6	\$7.5	87.9	88.7	88.5	88.5			8).9	Oye
10.	87	86	86	87	89	89	89	90	90	88	87	37	Op
834	92	3 1	90	31	93	94	94	94	95	54.	92	92	o _{हर}

Dornal Temperature Range.

Monthly Characteristics.

Daily 3.5807 Minimum 3 Op Naximum 5 Of Average 86,8°F Minimum. 36 °F (OMG) bre 17 June Maximum 95 °F (1603 bre 16 June)

ALWAND REPTIPERT SHADE TELPERATURE

A. A	00.	0200	0400	c600	2802	1000	3.200	17,00	2,600	1800	3 000	150 - 42	time.
്ളക		82.2	79.4	79.6	93.4	99.7	102.5	1046	304.7	102.8	96. 7	90	ъp.
ι,Ω _o	80	77	70	70	86	88	<i>3</i> 0	93	97	94	35	82	ာ့ ၊
-30%.c	91	8 8	87	90	100	106	110	114	1.14.	130	104	97	C.M

Monthly Characteristics.

Maximum 124 OF (1400 hrs 5 June)
Maximum 124 OF (1400 hrs 28 June)

MAY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALMAND REPLYERY GROUND TEMPERATURE AT A DEPTH OF 4 FEST.

ine	00.	02 00	04:00	0600	0800			3400		1800	2000	22.00	Hrs.
fge,	83.9	85.1	83,1	35.0	83.3	83.8	84.9	85.0	8k-, 9	846	83.8	83.4	ပန္ကာ
in.	79.5	79	73	74.5	77.0	79.0	7 9 .5	79°5	<i>7</i> 9∘5	79 _° 5	79.5	79.5	OF
ak.	88	88	83	88	88	88	90	90	90	90	88	88	Op

Dadly Temperature Range.

Monthly Characteristics.

Average 2.10F Minimum 1.00p Maximum 7.5°F

Average 83.9 op Minimum 73 Maximus 90

ALUAND REFINERY SHADE TEXTERATURE

ime	00.	0260	9400	c 600	೦೮ಎ೦	1000	1200	14.00	1600	1800	2000	22.00	Hra
7 80 .	79.5	7 5.7	73.5	72.7	84.0	89.8	92.9	94.9	94.07		87.9		op
in.	a	60	59	56	62	62°5	68	72	74	74	69	67.5	ck
1 X .	92	87	84	9C	94	104	105	104	104	103	98	9 8	O.Ju

Monthly Characteristics.

Average 85.17F
Minimum 56 °P (0600 hrs 16 May)
Maximum 105 °F (1200 hrs 30 May)

77.

AFRIL 1947.

STATISTICAL RESULTS

Imperature at 2 Hourly Intervals.

CLEAR BOXINERY GROUND TO MERATURE AT A DEPTH OF & FEET

<u> </u>	N.	0200	2400	96 66	0800	1000	1200	1400	1600	1800	2000	2 2.00	Hrs.
top.	16.2	16.1	76.3	76.5	76.5	76.6	76.7	76.7	76.E	76.3	76.0	75.8	o _F ,
, ·	12 1	12	74	72	70	7e.5	л.5	71.5	71.5	72.5	71.5	72	ok
	73	79	79-5	19.5	, 17 9 :	30	80	80.	79.5	79.5	79	79	ਹਲੂ

er er Comperationne America

Monthly Characteristics.

1.90g Tanaman ,50F Tanaman 3.50F Average 76.30F Minimum 70 °F (0800 bre 1 Apl) Maximum 80 °F (1200 bre 23 Apl)

ALE AND REFINERY SHADE TEMPERATURE

» 00 0200 0400 0600 0800 1000 1200 1400 1600 1800 2000 22c	O lies.
and the same of th	
	à
70. 60.5 6k.1 61.6 62.1 73.7 81.3 85.1 86.1 85.0 82.8 76.3 70.	977
7 54 59 50 50 50 64 69.5 70 71 70 65 60	op
at 36 93 66 84 46 97 98.5 39 96 94 92 36.	e e

Monthly Themsoteristics.

Ave. 604

Finding 90 97 (Jaco hrs 18 Apl)
haring 99 97 (Jl/O ars 26 Apl)

MARCH 1947

STATISTICAL RESULTS

Temporature at 2 Hourly Intervals.

ALWAND REPINERY CHOULD TELLERATURE AS A DESTRICT A WEST

Pune	00.	0860	9400	0600	0800	1000	1200	1400	1600	2800	2000	22.00	Has
Avge.	59.0	68 ₀ 8	68.7	68.6	69.0	69. 2	69.3	696	69.4	593	69.1	୫୨.୯	OŢ:
Man.	ő6 ₊ 5	66	66	66	66.5	66.5	67	67	G7.5	67.5	67	67	ch
Maz.	12	72	71.5	71.5	71	72	72	72	71.5	72.5	72.5	72.5	Oja

Daily Temperature Range.

Monthly Characteristics.

Average 1.807 Endmum 0.507 Endmum 5.007

vacuade Quote Minimum 66° (0200 hrs 1 March) Meximum 72° (1000 hrs 19 March)

ALMAND PERIMERY SHADE TENTERATURE

		-			Avenue at a villa	METER OF BUILDING	programme	***				to some an	* 1 Mar - 17	
	Time	0 0,	0200	ortoo	0600	030 0	.,	1200		1600 ·	į	2000	22 00	Ħ
•	A.ge	59.3	57.8	56.4	55.1	60.5		•			•	63.0	62.5	Opt
	llin.	52	50	&B	46	50	55	60-5	אנד	58	58	52	55	O/a
	Mare	72	73-5	74	74	70	86	86	86		E.4.	16	[19 3	¹ о м 1

Monthly Characteristics.

Awarage 63.99F Himmum 56 9F (0600 bml 0 Ferroll) Maximum 86 9F (1200 bms 22 March)

PRERUARY 1947.

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REFINERY GROUND TEMPERATURE AT A DEPTH OF A MEET

P _{AIN}	00.	0200	0400	0600	0800	7000	7500	1400	2600	1800	3060	22, 00	Acs.
æ.	65.7	66,5	66.5	65.5	5 6 6	66 _° 8	67.0	67.2	67.4	67.4x	67.0	66. 9	ON
.in.	66 .	66	66	66	66	66	665	66.5	66.5	66	S6	66	ୁଦ
3286	68	67.5	67.5	67.5	67.5	68	68	68	68.5	68	58	68.5	Opē

Paily Temperature Range.

Monthly Characteristics.

Average 1.20p Vinimum .5°F Kerimum 2.0°F Average 66.6°F Minimum 66 °F (0000 hrs 10 Feb) Haximum 68.5°F (1600 hrs 2 Feb)

ALMAND SHADE TE ESPATURE

ine	cc.	0200	0400	0600	6800	1000	1200	1400	1660	1800	2006	22.00	ıÈu.
wge.	49.2	47.6	46.3	44.6	47.6	53.4	58 .8	61.8	61.8	58-1	54.1	51-7	Ok.
iin.	39	36	孤	32.5	34	40	44	45	47	43	Ł.L	40	 ପଞ୍ଜ
DE.	60	57	57	55	55	63	69.5	72.5	1/4.	70	63	52.5	og-

Monthly Characteristics

Average 48.80F Minimum 32.5°F (0600 hrs 5 Feb) Maximum 74 °F (1600 hrs 15 Feb)

JANUARY 1947

STATISTICAL RESULTS

Temperature en 2 Hourly Intervals.

LANGED REFERENCE GROUND THE EXPATURE AT A DEEPTH OF A FEET

	. Trid	ου. :	0200	0400	0600	08 60	1000	3200	1400	1600	1800	2000	22,00	Hrs
i	i ger	69. D	58, 9	68.8	68,6	68.4	68.7	68.9	69,2	69.5	69.4	69.3	69.7.	op
:	Longia	6,7	97	67	67	67	ં દ	57.5	68	6 8	68	67.5	67	OE1
į	Now.	12	71.5	70.5	70	70	74	72.5	71	72	72	72	72.5	op

Dualy Temperature Range.

Monthly Characteristics.

Average 2,009 Francis 2,009 Formass 5,009

Average 68.90F Einimum 67 of (0800 hrs 4 Jan.) Haximum 72 of (1600 hrs 4 Jan.)

ALMAID SCHUERT SPADE TENFERATURE

i I Ter ,	G (2)	9200	0400	06 00	03 00	2000	1200	2200	i į	1800	1		Hre
ge.	517	51.1	50, 1	50.2	51.1	55.1	581	60°E	60, k	57.3	54.5	52.2	oF
: ::::,	43	43	¥0	36	5 6	43	50	53	52	<i>1</i> ₂ 9	£4.	44	OF.
	, 5% 5%		פינד	, 62 1	66	GI,	 63	50	58	64.	68	60	Oğ.

Monthly Characteristics

Arrege 54 tor Minimum 36 or (0500 hrs 26 Jan.) Maximus 59 or (1400 hrs 13 Jan.)

17/31 DECEMBER 1946

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REFERENCE GROUND THE PERATURE AT A DEPTH OF L. PEET.

inc	00 á	0200	0400	0600	0860	1000	1200	1400			2000	22.00	Hrs.
gs.	72.9	73.0	72.8	72. ₀ 6	72.5	72.5	72.5	7 5.0	73.4	73.6	73.4	73.2	o _E
3 13 .	71	70.5	70.5	70.0	70	69.8	69	69	772	72.5	72.5	72.0	€ T F
:X.	76	7 7	76.5	76	75	75	75	76	76	77	77	77	ob

Diurnal Temperature Range.

Monthly Characteristics.

Average 2.20F Minimum 1.00F Maximum 4.00F Average 72.90F Minimum 59 °F (1200 lrs 31 Dec.) Meximum 77 °F (0200 hrc 18 Dec.)

ALWAND METINERY CHADE THE TERATURE

					·		E. ST. W. 171.	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	مداد المساهيين		PROGRAMME THE PERSON	BELLEY 1784 MORE THATAL	
me	00 _°	92 00	O4.00	060 0	0800		1200	1400	,	1800		1	Ile s.
£8.	48.0	46.2	45.1	44.6	44.6				!				op
31.	35°0	34.02	32.5	31.5	3 0.6	39.0	49.0	53.0	50 0	47.2	 42.0	40.e	o _F
M .	59.0	59.0	58.0	57.8	58.0	64 , 0	74c0	71.c	70°C	62.0	59.0	58.0	्रजूर इं
	·	-	J		<u> </u>	1					Married Control of the Control	America	,

Monthly Characteristics.

Average 50.000 | (0800 hts 29 Decs) | Naximum 74.000 (1200 bts 30 Decs)

(7) OTH TELFTERATURES AT AMADAM, TRUE

: Aget o

NOVEMBER 1947

STATISTICAL RESULTS

Tempost, are at 2 Hoursty intervals.

ABADAN DICCHING GRUDE THE SERATURES

(SURFACE LINE OFSTER)

Δŧ	63.	stor.	ça,ch	3600	once	1000	1250	ucc.	1600	3.800	2000	22.00	Hrs.
ge.	57 a	به جر ا ج	Sign of	F6 5	58,2	59	ар. _{Т.}	sō ⁴	34	76.6	79.1	65.4	O.e
	0												35
٠,	7]			73	13	. 95	112	136	104	100	29	81	C)

richy Comporatione Came.

Moraldy Cheracteristics.

বি গণস্থাক সূত্র করিছা তেওকালেলা ত তথি স্থাপন্তার কুটি তথ্

- Average - 68.548 - Mirikum - 69 - 77 - Makarum - 37.6-28

Aplant and the shorten

	i Tru	0200	0420	, 0600	080.	.£00k	3204	3400 i	1.500 i	uson	2000	Commence of the commence of th	Ars
,,,	57.5	65. :	Car in	62.8	64. G	72.5	رد. 16	90.2	Sec 4	75.2	71.5	69	Copp
				52		1	1 1		. 1	1			op.
S X	1 75.5	80	30 ¹	80	F.	88	56	95	45 -	i g v	83.5	es ·	Capa :

Comprehensive form 12 commercial action control action and a section of the commercial action and a section of the commercial action and action and action a

enderseiche der Sieder Reinfelder der Sieder Siederseiche der Sieder

84.

OCTOBER 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ABADAN INCOLLEG CRUDE TE PERATURES

(SURFACE LINE SYSTEM)

ine	00.	0200	2400	0600	08 00	1000	1200	3.4±00	1600	1800	2000	22,00	Ers.
age.	777	73.1	69.6	67 <u>°</u> 0	68.8	86.1	102.5	111.4	112.8	103.8	92.0	83.7	ok
in.	67	62	58	56	58	59	74	88	100	90	80	72	O.
az.	86	82	76	73	83	308	115	125	128	126	210	96	CF1

Daily Temperature Range.

Monthly Characteristics.

Average 46,10F
Minimum 34 °F October 24th
Maximum 57 °F October 3nd

Average 87.409 Minimum 56 99 Maximum 108 97

ABADAN SHADE TELFERATURES

ine	00.	0200	0400	0600	58 00	1.000	1260	1400		1000	2005	21. 00	Heb.
nge.	77.3	75.5	74.3	72.2	76.0	89,2	97.5	100,6	93 4	34.9	8(i	80.00	i Laga I
² D _c	69	66	64.5	63	6ê	76	8 7)	3 0		30	TV :	72	Typ.
ex.	85	83	82	31.5	38	1.02	309	312	112	105	94	δr.	e1

in some minimum in the second

Average 84.8°F Minimum 53 °F ASSESSES ALC TO CLASS FOR COURSE

SKITTE-BER 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ABADAI INCOMING CRUDE TRUERATURES

(SURPACE LINE SYSTEM)

Timo	62 76	0 %00	céno	റദ്രാ	3000	1.200	1.00	1/00	1800	2000	2700	24,500	Hrs.
Avge.	78.9	76.0	73.0	73.7	98.3	13.5.6	120,6	12006]		89.0	83.2	O-Jp
Min.	74.	71	68	71	90	107	115	111	200	9 0	81	80	op
Nax.	8£.	82	82	87	!			1	320	106	97	88	OP*

Daily Temperature Range.

Monthly Characteristics.

Average 47.6°F Minimum 39 Marimum 54

Average 94.95°P Minimum 68 °P Naximum 132 of

ABADAN SHADE TELEBRATURES

		-	-	ew during fact;	To be a street of the state of	Visitin dillion and La	na e wemain'i Lacatent	776-E-S-8 NB-44-341-4	THE ASSESSMENT OF THE PARTY OF	and an arrangement	THE SERVICE TO	and or works.	
Pinc	co.	0200	04.0 0	o600	0800	•	1			1800	1 1		Bre.
Avge.	85.7	83.1	807	78.3	82.8				1		1	;	ùy.
Mn.	7E	76	73.5	71	74	89	97	99	99	94	86	82	op
Xaxo	.93	90	89	86	90	3.04	114	115	115	108	97	54.	

Monthly Characteristics

Average 91.7°F Minimum 71 °F (0600 hrs 21 Sept.) Maximum 115 °F (1400 hrs 13 Sept.)

29.594 29.594

spitales Claims care and alling

	nger spacing	Same series comments	-	Services of the Sale Services	e autoreres	-		garage de la compansión d La compansión de la compa		uni contra e	12 - 1	:	
చేస	E TOT	EOI	err	S°LTT	LTT.	977	orr	ror	76	96	ନ୍ଧ	60	į
Æ,	5-05	76	86	5 500	Spr	90£	#5	76	20	} 1 ₩ <u>[</u> }	£.,	Gb	, .
ď.	8° 5 6	8°66	*°LOT	उ ⁻ रशः	375°F	770°°	E.EO.E	5.59	3Yc	E.M.	# * ! ; ;	6 26	- 65.
oe XI.	.52°00	2002	3300	0097	ontri	- 998t	000X	0081	1305	and the second of the second o	parter for the first	The second	powerstruck

THUTARIET SOME MACARA

Average 102.60p Hintmin 75 vg (9600 hrs & Aug.) Hintmin 79 vg (1000 hrs 29 Aug.) Eddich Sparent Ev of Apparent Ac 66 Apparent

Monthly Characterizatios

the its Componentiate Rouge

	CHARLES AND		A DESCRIPTION OF THE PERSON OF	A STATE OF THE PARTY OF THE PAR			Ę.	:					
đ _o	26	SOL	TSO	şξŢ	6 £ £	95T	727	ETT	gor	18	6 9	\$6	. 4
ą.	ተፀ	7.6	001	ETE.	221 ,	757	921	16	75	<i>§!</i>	6/	(4g	
Ã,	0° z 6	o ୫6	0 404	413SI	9'827	S'att	1513	S-507	/ শূৰ	TH:	. 48		
.A7H	CC.42S	0000	0004	οοειτ	OOSA	Lat's	60aT	. 50f c	00%	0000	laces Laces	(603) (1200)	1 .502
	6-32 magazina spec	Kamerus, I rawn	Approved the contract			AND HER PERSON AND ADDRESS OF THE PERSON A	- 		.,				

(SIMBYOR THE EXELEN)

SAMPLY NOT THE CROSS TE PERATURE

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STATESTICAL PUSHINGS

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87.

CONTROTTICAL RESULTS

THE ! 1957

Temperature at 2 Hourly intervals.

ABADAN LICELING CHUPS IN IRATURES

(SURPADE LIGH SYSTEM)

*	' ∢ :	200	04:00	0600	C806	2000	7500	14 20	2600	1800	2000	2200	2400	15 21
	- -	ā. ?	37.3	8 ‰€	90,6	2 / 10. 5	124.5	1,52.8	1]%=0	226,5	332.0	102.6	55.5	oğ.
		: · ·	33	78	43	e sera P (315	125	127	126	93 .	96	92	0င္င
٠,		is i	97	93	99	7.F3.	3.50	142	14.7	141.	1.28	1.16	7017	Q.6

Pally "emperature Range

Monthly Characteristics

Averago 50,9°F Minimum 39 °C Paranam 6,7 °F

Average 107 Jer Minimum 78 9r Maximum 127 9r

ABADMI MRADE TEMPERATURE

.3	•• ·	0000	0 290	0400	060.7	0800	1000	1200	1400	3500	1/00	2-200	5300	No a
g.,	;	地方	92.2	20.1	88 .5	95-5	1.06.0	112.7	115.2	114,09	110 a	10 1 . ?	97.3	(3)p
*)	:	89.5	84.	8 4 -	82	91	90	107	3.2.2.	170	100) ! !	92	Ť.
	į,	200	98	95	94.	1.0k.	116	119	123	1°4	119	2.09	10%	oğ.

Invitin Characteristics

Traisem 32 ^D (2600 12 July)

Traisem 124 ^O (3600 hear) 3 July

STATISTICAL RESULTS

JUNE 1947

Temperature at ? Hourly Intervals

ABADAN INCOMING CRUDE TIMERATURES

(SURPACE LINE SESTEM)

ire	0200	0400	060C	රුපුර	1000	1290	1400		1800	2000	2200	2400	Hr.
rge.	86.5	82.8	79.4	88.8	30%7			1		109.8			3/4
in.	82	78	75	76	94	105	118	129	313	3.01	91	86	o _{li}
'ax.	95	89	84	39	115	130	142	145	143	131	134	1.05	्रमु

Diurnal Temperature Rango

Monthly Characteristics

Average 52.50F Minimum

Average 104,4°F Minimum 75 °F (0600 Hrs. 6 June) Maximum 145 °F (1600 Hrs. 28 June)

ARADAN SHADE TEMPERATURES

		-			marada tira i dili an-	paret Met. Per it	T May all street		p aradi 1 1110	-		richida e estada inte	*:=!:::::	
	inc	0000	0200	0400	06 00	9 800	i	Ì		2.600	3800	2000	2200	Hys.
	rge.	91.1	89.1	36.4	34.8		2.06.2			111.9	207.8	98.4	95.9	Dja
j	în.	86	84	79	76	87	96	99	3.00	1.00	98	92	88.5	O.B.
	2 3 6.	97	94	92	92	_	lli:	119	120	121.5	127	105	99	o'p

Monthly Characteristics

Average 99.0°F Minimum 76.0°F (0600 brs 6 June) Maximum 121.5°F (1600 brs 28 June)

MAY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ABADAN INCOMING CRUDS TRAFFRATURES

(SURPLOK LINE SYSTEM)

	1700	OL/10	0600	C80C	1000	1200	1400	1500	3800	2000	2200	2400	lie s
٠,	N2 2	76.3	75.3	76 .5	90.9	108.2	119.9	124,2	120.7	108.7	96.1	84, 4.	هر
:	, t.b.	<i>€</i> 5	ड्रन	63	72	90	101	101	3.00	9 0	84.	75	op
£ + ,	9Ŀ		!	1	104	121	134	139	137	126	110	20a	غد

10 Magicine Care Phage

Monthly Characteristics

1977 ang 1975 (1975) 1978 ang 1975 (1975) 1978 ang 1975 (1975)

Average 104.5°F Menimum 62 °F (0600 hrs 1st May) Meximum 139 °F (1600 hrs 27th May)

ABADAM STADE TEMPERATURE

		•	3060			1200	140C	3.600	3800	2000	2200	Hrs.
47 g	81. 1	79.5	78.2	87.2	96.5	ئ _ە 101	103.5	102.6	99,2	92.4	86° €	^#
ro j	. 7a	68	66	76	3 3.	85	87	86	82	77	75	्रक
) 	92	90	88	leo	170	1.14	118	115	113	203	97	(-Jh

Harriday Glamma berindles

Average 90.89P Maintage 66 99 (0600 hmw 16 Wey)

APRIL 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ABADAN INCOMING CRUDE TEMPERATURES

(SUBFACE LINE SYSTEM)

	M.	0200	04:10	0600	0800	1000	1200	14,00	1	1800	2000		2400	
•	75.6	75.7	69.4	66.4	67 ,2	77.1	97.0	103.6				!	ì	ì
	5.R	58	56	5 5	55	64-	78	92	93	87	85	72	68	Op
	85	87	78	75	105	306	1.15	123	127	124.	110	99	90	o _F

immel Temperature Range

Monthly Characteristics

Average 47.3°F Mindmum 28 °F Morimum 60 °F

Average 85.7°F Minimum 55 °F (0600 hrs 19 April) Moximum 127 °F (1600 hrs 25 April)

BADAN SHADE TEMPERATURES

Iro	0000	0200	0400	0600	0800	1000	1200	1400	1600	1800	2000	2200	Fæs.
Tge.	75.1	72.0	69.9	68.0	74.8	85.↓	91.4	93.6	93.4	89.8	82.8	78.8	estin Estin
in.	64	62	61.5	57	66	75	77	77	76	74	70	67	op
ar,	85	83	80	79	83	97	102.5	103.5	104	98	93	68) De

Monthly Characteristics

Average 81.25°F
Minimum 57 °F (0600 hrs 23 April)
Maximum 104 °F (1600 hrs 25 April)

STATISTICAL RESULTS

MARCH 1947

Temperature at 2 Hourly intervals.

ARADAM INCOMING CRUDE TEMPERATURES

(SURFACE LINE SYSTEM)

*********	-							,			CANADA AND		
ire	1	0200	OHCC	0600	0300	1000	1200	1400	1600	1800	2000	2200	Hrs.
.vge.	65.0	62.0	59.4	58.1	63.4	77.7	90.0	95.6	91.2	86.8	76.9	70 _. 5	င့္ခ
iin.	60	56.0	54	51.	45	<i>5</i> 7	70	65	6 6	67	65	63	ow
az.	72	70	67	65	80	9 7	108	109	104.	96	88	76	c _F

Diurnal Temperatures Range

Monthly Characteristics

Average 39.10F Mintenn 5 or Maximum 48 or Average 74.7°F Minimum 49 °F (0800 hrs 2 March) Maximum 109 °F (1400 hrs 27 March)

ABADAN SHADE TEMPERATURES

ine	0000	0200	0400	0600	0800	1000	1200	34,00		1800	2000	2200	Hrs,
in.	61.5	•	63.3 56 70	61.5 54 69	64.2 54 75	71.5 62 82	77。8 73 87	l	80.5 69.5 92	1	72.4 66 81	69.0 65 78	طَ ن م لَّد تأت

Monthly Characteristics

Average 70.8°F Minimum 54 °F (0800 hrs 2 Merch) Maximum 92 °F (1600 hrs 28 March)

92.

STATISTICAL RESULTS

FRERUARY 1947

Temperature at 2 Hourly intervals

ABADAN INCOMING CINIDS TENTERATURES

(SURFACE LINE SYSTEM)

leng.	0000	0200	01400		0800	1000	1200	1400	1600	1800	2000	2200	Hirs
:60±	54.7	51.2	:	2	45°2	57.2	69.7	77.4	79.4	7406	65.8	59.5	Ofe
i.	16	43	39	37	36	42	56	57	57	55	56	52	Og.
re	6 6	63	60	59	59	66	80	87	91	88	77	72	ာင္နာ

Diurnal Temperature Range

Monthly Characteristics

Average 33,60m Average 5 op Meximum 50 op

Average 60.95 F
Minimum 36 F (0800 hrs 7 Feb.)
Haximum 91 F (1600 hrs 17 Feb.)

ABADAN SHADE TEMPERATURES

imo	9000	0200	04.60	0600	0 800	1000	1200	1400	1600	1800	2000		Ere.
1 3 9-	56.5	54.8	53.2	51.8	51.6	58.7	65.3	67.8	68.7	66.1	61.8	58.9	o _{ľ,}
in	49	48	46	late.	4.5	50	56	58	58.5	57	55	515	ीम
ex.	63	64	62	61.5	63	66	7h	76	76	72	69	68	OP.

Monthly Characteristics

Average 59.6°F
Minimum 45 °F (0800 hrs 7 Feb.)
Naximum 76 °F (1400 hrs 27 Feb.)

JANUARI 1947

STATISTICAL RESULTS

Temperature at 2 Hourly intervals

AHADAN INCOMING CRUDE TEMPERATURES

(SURFACE LINE SYSTEM)

											-	AND DESCRIPTION OF THE PERSON	
PAGE .	6000	0200	0400	0600	0800	7000	1200	1400	1600	1860	2000	2200	Erve
rge.	55.5	51,1	49.2	48.1	48.1	5 5.6	65.8	72.5		67.5		56.6	or
 .D.	i s ė.	<u>i.2</u>	40	39	40	47	56	60	60	59	50	46	O.
JL.	55	65	62.	60	60	69	78	82	Blip	80	70	64.	ale
		;	ì	}	1 :		I		And white the same	Commence of the last of the last	COMMENT STREET	ACTION TO THE	2

Reversal Temperature Range

Monthly Characteristics

iverage 25.5°B Arnimum 5 op Maximum 41 op Average 58.5°F Mindmum 59 °F (0600 hrs 27 Jan.) Maximum 84 °F (1600 hrs 12 Jan.)

ABADAY SHADE TEMPEDATURES

тре	0000	0200	0400	e 60 0	0800	1000	1200	1400	160 0	1800	2000	2200	Ers.
rge.	55. 9 50	1	1	52.1 46	52.2 46	58.4 51.		65.6 57	6 6.0	62.9 58	59.7 52		oj.
Œ.	65	63	64	64	67	70	72	73	73	69	67	1	OF.

Monthly Characteristics

Average 58.4°F
Minimum 46 °F (0400 hrs 18 Jan.)
Maximum 73 °F (1400 hrs 31 Jan.)

DECEMBER 1946

STATISTICAL RESULES

Temperature at 2 hourly intervals

ABADAN INCOMPRE CRUINE TEMPREATURES

(SUMPACE LINE STREET)

L		c 200	Ohoo	060 6	0600	1	1200		1	3.800	5000	2200	Hr.s.
8	545.5	51.8	50.3	48.5		(1			68.7	62.6	58.2	o _{le}
Ŧ	3.2	39	35	34	3 5	43	55	63	65	60	52	47	op.
ŭ.	65	60	60	59	59	65	78	Bi _t	87	80	78	70	υħ.

Enwari Temperature Renge

Average 21, 10F Minimum 14

Maximum 39 OF

Monthly Characteristics

Average 59,509 Minimum 33 °F (0800 hrs 24 Dec.) Maximum 87 °F (1600 hrs 1 Dec.)